

# Adipic acid, isohexyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C17H30O4/c1-4-5-8-13-20-16(18)11-6-7-12-17(19)21-14-9-10-15(2)3/h4,15H,1
<b>InchiKey:</b>	DQGXHUOVOKREPW-UHFFFAOYSA-N
<b>Formula:</b>	C17H30O4
<b>SMILES:</b>	C=CCCCOC(=O)CCCC(=O)OCCCC(C)C
<b>Mol. weight [g/mol]:</b>	298.42

## Physical Properties

Property code	Value	Unit	Source
gf	-290.18	kJ/mol	Joback Method
hf	-763.66	kJ/mol	Joback Method
hfus	40.56	kJ/mol	Joback Method
hvap	70.69	kJ/mol	Joback Method
log10ws	-4.28		Crippen Method
logp	4.036		Crippen Method
mvol	260.970	ml/mol	McGowan Method
pc	1363.65	kPa	Joback Method
rmpol	2012.00		NIST Webbook
rmpol	2012.00		NIST Webbook
tb	737.18	K	Joback Method
tc	917.39	K	Joback Method
tf	408.91	K	Joback Method
vc	1.010	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	766.33	J/molxK	737.18	Joback Method
cpg	840.98	J/molxK	887.36	Joback Method
cpg	827.74	J/molxK	857.32	Joback Method
cpg	813.66	J/molxK	827.29	Joback Method
cpg	798.75	J/molxK	797.25	Joback Method
cpg	782.97	J/molxK	767.22	Joback Method
cpg	853.40	J/molxK	917.39	Joback Method
dvisc	0.0000793	Paxs	737.18	Joback Method

dvisc	0.0001054	Paxs	682.47	Joback Method
dvisc	0.0001473	Paxs	627.76	Joback Method
dvisc	0.0002194	Paxs	573.04	Joback Method
dvisc	0.0003555	Paxs	518.33	Joback Method
dvisc	0.0006455	Paxs	463.62	Joback Method
dvisc	0.0013750	Paxs	408.91	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353794&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353794&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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