

# Diethylmalonic acid, monochloride, 2-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C13H14Cl2O3/c1-3-13(4-2,11(15)16)12(17)18-10-8-6-5-7-9(10)14/h5-8H,3-4H
<b>InchiKey:</b>	NNXBFWGWMZXTHU-UHFFFAOYSA-N
<b>Formula:</b>	C13H14Cl2O3
<b>SMILES:</b>	CCC(CC)(C(=O)Cl)C(=O)Oc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	289.15

## Physical Properties

Property code	Value	Unit	Source
gf	-222.50	kJ/mol	Joback Method
hf	-484.20	kJ/mol	Joback Method
hfus	28.44	kJ/mol	Joback Method
hvap	70.85	kJ/mol	Joback Method
log10ws	-4.25		Crippen Method
logp	3.817		Crippen Method
mcvol	203.760	ml/mol	McGowan Method
pc	2276.24	kPa	Joback Method
rinpol	1820.00		NIST Webbook
rinpol	1820.00		NIST Webbook
tb	730.29	K	Joback Method
tc	958.68	K	Joback Method
tf	459.56	K	Joback Method
vc	0.772	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	514.54	J/molxK	730.29	Joback Method
cpg	567.14	J/molxK	920.61	Joback Method
cpg	558.38	J/molxK	882.55	Joback Method
cpg	548.79	J/molxK	844.48	Joback Method
cpg	538.33	J/molxK	806.42	Joback Method
cpg	526.93	J/molxK	768.35	Joback Method
cpg	575.13	J/molxK	958.68	Joback Method
dvisc	0.0001180	Paxs	730.29	Joback Method

dvisc	0.0001507	Paxs	685.17	Joback Method
dvisc	0.0001992	Paxs	640.05	Joback Method
dvisc	0.0002748	Paxs	594.92	Joback Method
dvisc	0.0003996	Paxs	549.80	Joback Method
dvisc	0.0006213	Paxs	504.68	Joback Method
dvisc	0.0010535	Paxs	459.56	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369630&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369630&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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