

# Hexachlorophene dimethyl ether

<b>Inchi:</b>	InChI=1S/C15H11Cl5O2/c1-21-12-4-3-9(16)13(19)7(12)5-8-14(20)10(17)6-11(18)15(8)2
<b>InchiKey:</b>	ZZFFLZBDRDHFBT-UHFFFAOYSA-N
<b>Formula:</b>	C15H11Cl5O2
<b>SMILES:</b>	COc1ccc(Cl)c(Cl)c1Cc1c(Cl)c(Cl)cc(Cl)c1OC
<b>Mol. weight [g/mol]:</b>	400.51

## Physical Properties

Property code	Value	Unit	Source
gf	-36.82	kJ/mol	Joback Method
hf	-303.30	kJ/mol	Joback Method
hfus	43.33	kJ/mol	Joback Method
hvap	84.91	kJ/mol	Joback Method
log10ws	-7.24		Crippen Method
logp	6.562		Crippen Method
mvol	247.630	ml/mol	McGowan Method
pc	1857.91	kPa	Joback Method
rinpol	2666.00		NIST Webbook
tb	862.81	K	Joback Method
tc	1108.45	K	Joback Method
tf	593.35	K	Joback Method
vc	0.941	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	583.83	J/molxK	862.81	Joback Method
cpg	623.17	J/molxK	1067.51	Joback Method
cpg	617.49	J/molxK	1026.57	Joback Method
cpg	610.71	J/molxK	985.63	Joback Method
cpg	602.83	J/molxK	944.69	Joback Method
cpg	593.87	J/molxK	903.75	Joback Method
cpg	627.74	J/molxK	1108.45	Joback Method
dvisc	0.0000643	Paxs	862.81	Joback Method
dvisc	0.0000755	Paxs	817.90	Joback Method

dvisc	0.0000904	Paxs	772.99	Joback Method
dvisc	0.0001107	Paxs	728.08	Joback Method
dvisc	0.0001391	Paxs	683.17	Joback Method
dvisc	0.0001805	Paxs	638.26	Joback Method
dvisc	0.0002437	Paxs	593.35	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R543152&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R543152&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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