

# Propoxylated neopentyl glycol diacrylate

<b>Inchi:</b>	InChI=1S/C14H22O5/c1-6-12(15)18-10-14(4,5)9-17-8-11(3)19-13(16)7-2/h6-7,11H,1-2,8
<b>InchiKey:</b>	SDMYEUPNWFYNGW-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O5
<b>SMILES:</b>	C=CC(=O)OCC(C)(C)COCC(C)OC(=O)C=C
<b>Mol. weight [g/mol]:</b>	270.32

## Physical Properties

Property code	Value	Unit	Source
gf	-329.76	kJ/mol	Joback Method
hf	-717.28	kJ/mol	Joback Method
hfus	25.28	kJ/mol	Joback Method
hvap	64.46	kJ/mol	Joback Method
log10ws	-2.07		Crippen Method
logp	1.876		Crippen Method
mvol	220.270	ml/mol	McGowan Method
pc	1771.36	kPa	Joback Method
rinpol	1619.00		NIST Webbook
rinpol	1619.00		NIST Webbook
tb	684.41	K	Joback Method
tc	874.80	K	Joback Method
tf	397.99	K	Joback Method
vc	0.831	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	606.66	J/molxK	684.41	Joback Method
cpg	621.66	J/molxK	716.14	Joback Method
cpg	635.81	J/molxK	747.87	Joback Method
cpg	649.14	J/molxK	779.61	Joback Method
cpg	661.65	J/molxK	811.34	Joback Method
cpg	673.37	J/molxK	843.07	Joback Method
cpg	684.31	J/molxK	874.80	Joback Method
dvisc	0.0012803	Paxs	397.99	Joback Method

dvisc	0.0006300	Paxs	445.73	Joback Method
dvisc	0.0003556	Paxs	493.46	Joback Method
dvisc	0.0002220	Paxs	541.20	Joback Method
dvisc	0.0001496	Paxs	588.94	Joback Method
dvisc	0.0001070	Paxs	636.67	Joback Method
dvisc	0.0000801	Paxs	684.41	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R508540&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R508540&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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