

# «beta»-Isopropoxystyrene, (E)

Inchi:	InChI=1S/C11H14O/c1-10(2)12-9-8-11-6-4-3-5-7-11/h3-10H,1-2H3/b9-8+
InchiKey:	VNOSMVIVSUHSPU-CMDGGOBGSA-N
Formula:	C11H14O
SMILES:	CC(C)OC=Cc1ccccc1
Mol. weight [g/mol]:	162.23

## Physical Properties

Property code	Value	Unit	Source
gf	126.93	kJ/mol	Joback Method
hf	-54.12	kJ/mol	Joback Method
hfus	16.15	kJ/mol	Joback Method
hvap	44.34	kJ/mol	Joback Method
log10ws	-3.24		Crippen Method
logp	3.082		Crippen Method
mcvol	143.660	ml/mol	McGowan Method
pc	2770.08	kPa	Joback Method
rinpol	1274.00		NIST Webbook
rinpol	1274.00		NIST Webbook
tb	503.90	K	Joback Method
tc	718.62	K	Joback Method
tf	242.30	K	Joback Method
vc	0.535	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.42	J/mol×K	503.90	Joback Method
cpg	324.08	J/mol×K	539.69	Joback Method
cpg	338.82	J/mol×K	575.47	Joback Method
cpg	352.68	J/mol×K	611.26	Joback Method
cpg	365.69	J/mol×K	647.05	Joback Method
cpg	377.90	J/mol×K	682.83	Joback Method
cpg	389.34	J/mol×K	718.62	Joback Method
dvisc	0.0037869	Paxs	242.30	Joback Method

dvisc	0.0014587	Paxs	285.90	Joback Method
dvisc	0.0007232	Paxs	329.50	Joback Method
dvisc	0.0004225	Paxs	373.10	Joback Method
dvisc	0.0002762	Paxs	416.70	Joback Method
dvisc	0.0001957	Paxs	460.30	Joback Method
dvisc	0.0001472	Paxs	503.90	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R318104&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R318104&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/37-887-7/beta-Isopropoxystyrene-E.pdf>

Generated by Cheméo on 2024-04-19 18:57:00.053394191 +0000 UTC m=+15842268.973971507.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.