

# Dibutyl tartrate

<b>Other names:</b>	Dibutyl L-tartrate Dibutyl D-tartrate Butanedioic acid, 2,3-dihydroxy- [R-(R*,R*)]-, dibutyl ester
<b>Inchi:</b>	InChI=1S/C12H22O6/c1-3-5-7-17-11(15)9(13)10(14)12(16)18-8-6-4-2/h9-10,13-14H,3-8H
<b>InchiKey:</b>	PCYQQSKDZQTOQG-NXEZZACHSA-N
<b>Formula:</b>	C12H22O6
<b>SMILES:</b>	CCCCOC(=O)C(O)C(O)C(=O)OCCCC
<b>Mol. weight [g/mol]:</b>	262.30
<b>CAS:</b>	87-92-3

## Physical Properties

Property code	Value	Unit	Source
gf	-696.20	kJ/mol	Joback Method
hf	-1095.63	kJ/mol	Joback Method
hfus	33.54	kJ/mol	Joback Method
hvap	93.20	kJ/mol	Joback Method
log10ws	-1.32		Crippen Method
logp	0.395		Crippen Method
mcvol	206.560	ml/mol	McGowan Method
pc	2315.84	kPa	Joback Method
tb	810.02	K	Joback Method
tc	994.77	K	Joback Method
tf	460.96	K	Joback Method
vc	0.781	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	633.71	J/molxK	810.02	Joback Method
cpg	644.99	J/molxK	840.81	Joback Method
cpg	655.54	J/molxK	871.60	Joback Method
cpg	665.38	J/molxK	902.40	Joback Method
cpg	674.51	J/molxK	933.19	Joback Method
cpg	682.92	J/molxK	963.98	Joback Method

cpg	690.64	J/molxK	994.77	Joback Method
dvisc	0.0008016	Paxs	460.96	Joback Method
dvisc	0.0002009	Paxs	519.14	Joback Method
dvisc	0.0000666	Paxs	577.31	Joback Method
dvisc	0.0000270	Paxs	635.49	Joback Method
dvisc	0.0000127	Paxs	693.67	Joback Method
dvisc	0.0000067	Paxs	751.84	Joback Method
dvisc	0.0000039	Paxs	810.02	Joback Method
hvapt	79.80	kJ/mol	469.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	448.20	K	0.70	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C87923&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C87923&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/38-045-0/Dibutyl-tartrate.pdf>

Generated by Cheméo on 2024-04-19 16:45:15.851357798 +0000 UTC m=+15834364.771935113.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.