

# Diethylmalonic acid, ethyl 3-phenylpropyl ester

Inchi:	InChI=1S/C18H26O4/c1-4-18(5-2,16(19)21-6-3)17(20)22-14-10-13-15-11-8-7-9-12-15/h7
InchiKey:	POMWAAUAKUJFNG-UHFFFAOYSA-N
Formula:	C18H26O4
SMILES:	CCOC(=O)C(CC)(CC)C(=O)OCCc1ccccc1
Mol. weight [g/mol]:	306.40

## Physical Properties

Property code	Value	Unit	Source
gf	-251.91	kJ/mol	Joback Method
hf	-676.67	kJ/mol	Joback Method
hfus	34.58	kJ/mol	Joback Method
hvap	74.95	kJ/mol	Joback Method
log10ws	-3.94		Crippen Method
logp	3.532		Crippen Method
mcvol	255.600	ml/mol	McGowan Method
pc	1594.89	kPa	Joback Method
rinpol	2041.00		NIST Webbook
rinpol	2041.00		NIST Webbook
tb	787.27	K	Joback Method
tc	993.13	K	Joback Method
tf	465.78	K	Joback Method
vc	0.973	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	763.54	J/molxK	787.27	Joback Method
cpg	832.99	J/molxK	958.82	Joback Method
cpg	821.17	J/molxK	924.51	Joback Method
cpg	808.35	J/molxK	890.20	Joback Method
cpg	794.50	J/molxK	855.89	Joback Method
cpg	779.58	J/molxK	821.58	Joback Method
cpg	843.88	J/molxK	993.13	Joback Method
dvisc	0.0000577	Paxs	787.27	Joback Method

dvisc	0.0000763	Paxs	733.69	Joback Method
dvisc	0.0001055	Paxs	680.11	Joback Method
dvisc	0.0001542	Paxs	626.52	Joback Method
dvisc	0.0002418	Paxs	572.94	Joback Method
dvisc	0.0004162	Paxs	519.36	Joback Method
dvisc	0.0008118	Paxs	465.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369650&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369650&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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