

# Phthalic acid, pentadecyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C31H52O4/c1-6-7-8-9-10-11-12-13-14-15-16-17-20-23-34-29(32)27-21-18-19-
InchiKey:	NXOQIOKWSWIEMS-UHFFFAOYSA-N
Formula:	C31H52O4
SMILES:	CCCCCCCCCCCCCOC(=O)c1cccc1C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	488.74

## Physical Properties

Property code	Value	Unit	Source
gf	-154.52	kJ/mol	Joback Method
hf	-961.74	kJ/mol	Joback Method
hfus	64.34	kJ/mol	Joback Method
hvap	104.17	kJ/mol	Joback Method
log10ws	-10.26		Crippen Method
logp	9.164		Crippen Method
mvol	438.770	ml/mol	McGowan Method
pc	707.71	kPa	Joback Method
rinpol	3279.00		NIST Webbook
tb	1089.25	K	Joback Method
tc	1345.31	K	Joback Method
tf	609.81	K	Joback Method
vc	1.694	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1561.14	J/molxK	1089.25	Joback Method
cpg	1640.09	J/molxK	1302.63	Joback Method
cpg	1627.50	J/molxK	1259.96	Joback Method
cpg	1613.45	J/molxK	1217.28	Joback Method
cpg	1597.80	J/molxK	1174.60	Joback Method
cpg	1580.41	J/molxK	1131.93	Joback Method
cpg	1651.36	J/molxK	1345.31	Joback Method
dvisc	0.0000070	Paxs	1089.25	Joback Method
dvisc	0.0000095	Paxs	1009.34	Joback Method

dvisc	0.0000136	Paxs	929.44	Joback Method
dvisc	0.0000209	Paxs	849.53	Joback Method
dvisc	0.0000350	Paxs	769.62	Joback Method
dvisc	0.0000662	Paxs	689.72	Joback Method
dvisc	0.0001477	Paxs	609.81	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377781&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377781&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/38-267-4/Phthalic-acid-pentadecyl-2-4-4-trimethylpentyl-ester.pdf>

Generated by Cheméo on 2024-04-28 12:24:00.065111863 +0000 UTC m=+16596288.985689179.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.