

# Propanenitrile, 3-(diethylamino)-

<b>Other names:</b>	Propionitrile, 3-(diethylamino)- N-(2-Cyanoethyl)diethylamine 3-(Diethylamino)propionitrile 3-(Diethylamino)propanenitrile 3-diethylaminopropionitrile
<b>Inchi:</b>	InChI=1S/C7H14N2/c1-3-9(4-2)7-5-6-8/h3-5,7H2,1-2H3
<b>InchiKey:</b>	LFFKXGFSDGRFQA-UHFFFAOYSA-N
<b>Formula:</b>	C7H14N2
<b>SMILES:</b>	CCN(CC)CCC#N
<b>Mol. weight [g/mol]:</b>	126.20
<b>CAS:</b>	5351-04-2

## Physical Properties

Property code	Value	Unit	Source
gf	252.02	kJ/mol	Joback Method
hf	44.60	kJ/mol	Joback Method
hfus	18.41	kJ/mol	Joback Method
hvap	43.70	kJ/mol	Joback Method
log10ws	-1.19		Crippen Method
logp	1.242		Crippen Method
mcpvol	120.850	ml/mol	McGowan Method
pc	2721.17	kPa	Joback Method
tb	474.08	K	Joback Method
tc	659.05	K	Joback Method
tf	213.25 ± 0.30	K	NIST Webbook
tf	213.65 ± 0.20	K	NIST Webbook
vc	0.471	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	312.38	J/mol×K	628.22	Joback Method
cpg	260.17	J/mol×K	474.08	Joback Method
cpg	271.65	J/mol×K	504.91	Joback Method

cpg	282.59	J/mol×K	535.74	Joback Method
cpg	293.01	J/mol×K	566.57	Joback Method
cpg	302.94	J/mol×K	597.39	Joback Method
cpg	321.36	J/mol×K	659.05	Joback Method
hvapt	53.70	kJ/mol	404.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5351042&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5351042&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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