

# 3,6-octanedione

|                             |   |
|-----------------------------|---|
| <b>Inchi:</b>               | InChI=1S/C8H14O2/c1-3-7(9)5-6-8(10)4-2/h3-6H2,1-2H3 |
| <b>InchiKey:</b>            | CVZGUJMLZZTPKH-UHFFFAOYSA-N                         |
| <b>Formula:</b>             | C8H14O2   |
| <b>SMILES:</b>              | CCC(=O)CCC(=O)CC                                    |
| <b>Mol. weight [g/mol]:</b> | 142.20  |
| <b>CAS:</b>                 | 2955-65-9   |

## Physical Properties

| Property code | Value         | Unit                 | Source         |
|---------------|---------------|----------------------|----------------|
| gf            | -241.36       | kJ/mol               | Joback Method  |
| hf            | -433.61       | kJ/mol               | Joback Method  |
| hfus          | 19.67         | kJ/mol               | Joback Method  |
| hvap          | 46.89         | kJ/mol               | Joback Method  |
| log10ws       | -1.73         |                      | Crippen Method |
| logp          | 1.725         |                      | Crippen Method |
| mcvol         | 126.720       | ml/mol               | McGowan Method |
| pc            | 2890.51       | kPa                  | Joback Method  |
| ripol         | 1638.00       |                      | NIST Webbook   |
| tb            | 484.40 ± 2.00 | K                    | NIST Webbook   |
| tc            | 675.94        | K                    | Joback Method  |
| tf            | 279.78        | K                    | Joback Method  |
| vc            | 0.495         | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 275.20    | J/mol×K | 490.18          | Joback Method |
| cpg           | 286.97    | J/mol×K | 521.14          | Joback Method |
| cpg           | 298.23    | J/mol×K | 552.10          | Joback Method |
| cpg           | 308.97    | J/mol×K | 583.06          | Joback Method |
| cpg           | 319.22    | J/mol×K | 614.02          | Joback Method |
| cpg           | 328.98    | J/mol×K | 644.98          | Joback Method |
| cpg           | 338.27    | J/mol×K | 675.94          | Joback Method |
| dvisc         | 0.0036359 | Paxs    | 279.78          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0019569 | Paxs | 314.85 | Joback Method |
| dvisc | 0.0011925 | Paxs | 349.91 | Joback Method |
| dvisc | 0.0007953 | Paxs | 384.98 | Joback Method |
| dvisc | 0.0005675 | Paxs | 420.05 | Joback Method |
| dvisc | 0.0004266 | Paxs | 455.11 | Joback Method |
| dvisc | 0.0003340 | Paxs | 490.18 | Joback Method |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.42922e+01                   |
| Coeff. B                    | -3.82516e+03                  |
| Coeff. C                    | -6.75200e+01                  |
| Temperature range (K), min. | 340.66                        |
| Temperature range (K), max. | 493.45                        |

## Sources

|   |   |
|---|---|
| <b>Crippen Method:</b>                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>   |
| <b>Crippen Method:</b>                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>   |
| <b>Joback Method:</b>                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>McGowan Method:</b>                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>   |
| <b>NIST Webbook:</b>                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2955659&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2955659&amp;Units=SI</a>   |
| <b>The Yaws Handbook of Vapor Pressure:</b> | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |

## Legend

|                         |   |
|-------------------------|---|
| <b>cp<sub>g</sub>:</b>  | Ideal gas heat capacity                         |
| <b>dvisc:</b>           | Dynamic viscosity                               |
| <b>gf:</b>              | Standard Gibbs free energy of formation         |
| <b>hf:</b>              | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>            | Enthalpy of fusion at standard conditions       |
| <b>h<sub>vap</sub>:</b> | Enthalpy of vaporization at standard conditions |

|                 |                                     |
|-----------------|-------------------------------------|
| <b>log10ws:</b> | Log10 of Water solubility in mol/l  |
| <b>logp:</b>    | Octanol/Water partition coefficient |
| <b>mcvol:</b>   | McGowan's characteristic volume     |
| <b>pc:</b>      | Critical Pressure                   |
| <b>pvap:</b>    | Vapor pressure                      |
| <b>ripol:</b>   | Polar retention indices             |
| <b>tb:</b>      | Normal Boiling Point Temperature    |
| <b>tc:</b>      | Critical Temperature                |
| <b>tf:</b>      | Normal melting (fusion) point       |
| <b>vc:</b>      | Critical Volume                     |

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