

# Sebacic acid, butyl 3-phenylallyl ester

<b>Inchi:</b>	InChI=1S/C23H34O4/c1-2-3-19-26-22(24)17-11-6-4-5-7-12-18-23(25)27-20-13-16-21-14
<b>InchiKey:</b>	BQPJIXZFEXUBAZ-DTQAZKPQSA-N
<b>Formula:</b>	C23H34O4
<b>SMILES:</b>	CCCCOC(=O)CCCCCCCC(=O)OCC=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	374.51

## Physical Properties

Property code	Value	Unit	Source
gf	-132.43	kJ/mol	Joback Method
hf	-653.90	kJ/mol	Joback Method
hfus	55.14	kJ/mol	Joback Method
hvap	87.34	kJ/mol	Joback Method
log10ws	-6.30		Crippen Method
logp	5.707		Crippen Method
mcvol	321.750	ml/mol	McGowan Method
pc	1148.32	kPa	Joback Method
rinpola	2902.00		NIST Webbook
tb	909.06	K	Joback Method
tc	1116.49	K	Joback Method
tf	514.63	K	Joback Method
vc	1.244	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1031.09	J/molxK	909.06	Joback Method
cpg	1047.54	J/molxK	943.63	Joback Method
cpg	1062.81	J/molxK	978.20	Joback Method
cpg	1076.95	J/molxK	1012.77	Joback Method
cpg	1090.00	J/molxK	1047.34	Joback Method
cpg	1102.03	J/molxK	1081.91	Joback Method
cpg	1113.08	J/molxK	1116.49	Joback Method
dvisc	0.0004639	Paxs	514.63	Joback Method
dvisc	0.0002282	Paxs	580.37	Joback Method

dvisc	0.0001297	Paxs	646.11	Joback Method
dvisc	0.0000819	Paxs	711.85	Joback Method
dvisc	0.0000558	Paxs	777.58	Joback Method
dvisc	0.0000404	Paxs	843.32	Joback Method
dvisc	0.0000307	Paxs	909.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355889&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355889&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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