

# Propane, 1-(2-chloroethyl)-3-ethenyl

Inchi:	InChI=1S/C7H13ClO2/c1-2-9-5-3-6-10-7-4-8/h2H,1,3-7H2
InchiKey:	UVWCZJPMXMNGDG-UHFFFAOYSA-N
Formula:	C7H13ClO2
SMILES:	C=COCCCOCCCI
Mol. weight [g/mol]:	164.63

## Physical Properties

Property code	Value	Unit	Source
gf	-126.03	kJ/mol	Joback Method
hf	-342.56	kJ/mol	Joback Method
hfus	19.18	kJ/mol	Joback Method
hvap	39.71	kJ/mol	Joback Method
log10ws	-1.43		Crippen Method
logp	1.792		Crippen Method
mcvol	129.170	ml/mol	McGowan Method
pc	2715.50	kPa	Joback Method
rinsol	1648.00		NIST Webbook
tb	438.51	K	Joback Method
tc	613.65	K	Joback Method
tf	241.27	K	Joback Method
vc	0.493	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	254.26	J/molxK	438.51	Joback Method
cpg	264.79	J/molxK	467.70	Joback Method
cpg	274.99	J/molxK	496.89	Joback Method
cpg	284.86	J/molxK	526.08	Joback Method
cpg	294.40	J/molxK	555.27	Joback Method
cpg	303.60	J/molxK	584.46	Joback Method
cpg	312.46	J/molxK	613.65	Joback Method
dvisc	0.0025045	Paxs	241.27	Joback Method
dvisc	0.0012997	Paxs	274.14	Joback Method

dvisc	0.0007761	Paxs	307.02	Joback Method
dvisc	0.0005121	Paxs	339.89	Joback Method
dvisc	0.0003636	Paxs	372.76	Joback Method
dvisc	0.0002729	Paxs	405.64	Joback Method
dvisc	0.0002138	Paxs	438.51	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R502359&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R502359&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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