

# Glutaric acid, 3-ethylphenyl octyl ester

<b>Inchi:</b>	InChI=1S/C21H32O4/c1-3-5-6-7-8-9-16-24-20(22)14-11-15-21(23)25-19-13-10-12-18(4-2
<b>InchiKey:</b>	MTNBIMIQQZMFCG-UHFFFAOYSA-N
<b>Formula:</b>	C21H32O4
<b>SMILES:</b>	CCCCCCCCOC(=O)CCCC(=O)Oc1cccc(CC)c1
<b>Mol. weight [g/mol]:</b>	348.48

## Physical Properties

Property code	Value	Unit	Source
gf	-239.12	kJ/mol	Joback Method
hf	-741.31	kJ/mol	Joback Method
hfus	49.37	kJ/mol	Joback Method
hvap	83.59	kJ/mol	Joback Method
log10ws	-6.06		Crippen Method
logp	5.228		Crippen Method
mcvol	297.870	ml/mol	McGowan Method
pc	1252.15	kPa	Joback Method
rinpol	2607.00		NIST Webbook
rinpol	2607.00		NIST Webbook
tb	864.12	K	Joback Method
tc	1065.34	K	Joback Method
tf	509.69	K	Joback Method
vc	1.151	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	937.16	J/molxK	864.12	Joback Method
cpg	953.50	J/molxK	897.66	Joback Method
cpg	968.67	J/molxK	931.19	Joback Method
cpg	982.68	J/molxK	964.73	Joback Method
cpg	995.57	J/molxK	998.27	Joback Method
cpg	1007.37	J/molxK	1031.80	Joback Method
cpg	1018.08	J/molxK	1065.34	Joback Method
dvisc	0.0005353	Paxs	509.69	Joback Method

dvisc	0.0002901	Paxs	568.76	Joback Method
dvisc	0.0001764	Paxs	627.83	Joback Method
dvisc	0.0001169	Paxs	686.90	Joback Method
dvisc	0.0000826	Paxs	745.98	Joback Method
dvisc	0.0000615	Paxs	805.05	Joback Method
dvisc	0.0000476	Paxs	864.12	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359164&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359164&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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