

# Chloro(dimethyl)phosphine

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>2</sub> PCl Dimethyl chloro phosphine Phosphinous chloride, dimethyl-
<b>Inchi:</b>	InChI=1S/C2H6ClP/c1-4(2)3/h1-2H3
<b>InchiKey:</b>	ZLVVDNKTHWEIOG-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>2</sub> H <sub>6</sub> ClP
<b>SMILES:</b>	CP(C)Cl
<b>Mol. weight [g/mol]:</b>	96.50
<b>CAS:</b>	811-62-1

## Physical Properties

Property code	Value	Unit	Source
ie	9.00	eV	NIST Webbook
ie	8.70	eV	NIST Webbook
ie	9.30	eV	NIST Webbook
ie	9.20	eV	NIST Webbook
log10ws	2.34		Crippen Method
logp	1.882		Crippen Method
mcvol	71.740	ml/mol	McGowan Method
tf	272.00 ± 0.20	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	55.50	kJ/mol	250.50	NIST Webbook
hvapt	32.90	kJ/mol	289.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C811621&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C811621&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>tf:</b>	Normal melting (fusion) point

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