

# Diethylmalonic acid, decyl tetrahydrofurfuryl ester

Inchi:	InChI=1S/C22H40O5/c1-4-7-8-9-10-11-12-13-16-26-20(23)22(5-2,6-3)21(24)27-18-19-15
InchiKey:	BNPDQFIZHQJOHO-UHFFFAOYSA-N
Formula:	C22H40O5
SMILES:	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC1CCCO1
Mol. weight [g/mol]:	384.55

## Physical Properties

Property code	Value	Unit	Source
gf	-380.21	kJ/mol	Joback Method
hf	-1067.28	kJ/mol	Joback Method
hfus	52.81	kJ/mol	Joback Method
hvap	86.35	kJ/mol	Joback Method
log10ws	-5.61		Crippen Method
logp	5.199		Crippen Method
mcvol	330.730	ml/mol	McGowan Method
pc	1076.39	kPa	Joback Method
rinpol	2471.00		NIST Webbook
rinpol	2471.00		NIST Webbook
tb	894.34	K	Joback Method
tc	1097.66	K	Joback Method
tf	521.91	K	Joback Method
vc	1.266	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1120.54	J/molxK	894.34	Joback Method
cpg	1138.94	J/molxK	928.23	Joback Method
cpg	1156.03	J/molxK	962.11	Joback Method
cpg	1171.86	J/molxK	996.00	Joback Method
cpg	1186.49	J/molxK	1029.88	Joback Method
cpg	1199.97	J/molxK	1063.77	Joback Method
cpg	1212.34	J/molxK	1097.66	Joback Method
dvisc	0.0006388	Paxs	521.91	Joback Method

dvisc	0.0003134	Paxs	583.98	Joback Method
dvisc	0.0001763	Paxs	646.05	Joback Method
dvisc	0.0001097	Paxs	708.12	Joback Method
dvisc	0.0000737	Paxs	770.20	Joback Method
dvisc	0.0000525	Paxs	832.27	Joback Method
dvisc	0.0000392	Paxs	894.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370645&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370645&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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