

# Butanoic acid, 2-isopropyl-2,3,3-trimethyl, ethyl ester

<b>Other names:</b>	Butanoic acid, 2,3,3-trimethyl-2-(1-methylethyl), ethyl ester
<b>Inchi:</b>	InChI=1S/C12H24O2/c1-8-14-10(13)12(7,9(2)3)11(4,5)6/h9H,8H2,1-7H3
<b>InchiKey:</b>	AYKXSKJFLXRXGM-UHFFFAOYSA-N
<b>Formula:</b>	C12H24O2
<b>SMILES:</b>	CCOC(=O)C(C)(C(C)C)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	200.32

## Physical Properties

Property code	Value	Unit	Source
gf	-180.52	kJ/mol	Joback Method
hf	-558.59	kJ/mol	Joback Method
hfus	11.27	kJ/mol	Joback Method
hvap	48.48	kJ/mol	Joback Method
log10ws	-2.98		Crippen Method
logp	3.258		Crippen Method
mvol	187.380	ml/mol	McGowan Method
pc	1927.05	kPa	Joback Method
rinpol	1237.00		NIST Webbook
tb	543.35	K	Joback Method
tc	736.01	K	Joback Method
tf	287.00	K	Joback Method
vc	0.704	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	470.19	J/molxK	543.35	Joback Method
cpg	488.23	J/molxK	575.46	Joback Method
cpg	505.28	J/molxK	607.57	Joback Method
cpg	521.37	J/molxK	639.68	Joback Method
cpg	536.56	J/molxK	671.79	Joback Method
cpg	550.88	J/molxK	703.90	Joback Method
cpg	564.37	J/molxK	736.01	Joback Method
dvisc	0.0070187	Paxs	287.00	Joback Method

dvisc	0.0024540	Paxs	329.73	Joback Method
dvisc	0.0010919	Paxs	372.45	Joback Method
dvisc	0.0005740	Paxs	415.18	Joback Method
dvisc	0.0003402	Paxs	457.90	Joback Method
dvisc	0.0002205	Paxs	500.62	Joback Method
dvisc	0.0001530	Paxs	543.35	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R108195&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R108195&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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