

# 3-(2-Carboxy-ethyl)-oxirane-2-carboxylic acid methyl ester

Inchi:	InChI=1S/C7H10O5/c1-11-7(10)6-4(12-6)2-3-5(8)9/h4,6H,2-3H2,1H3,(H,8,9)
InchiKey:	TZJFDVISKBEVHZ-UHFFFAOYSA-N
Formula:	C7H10O5
SMILES:	COC(=O)C1OC1CCC(=O)O
Mol. weight [g/mol]:	174.15

## Physical Properties

Property code	Value	Unit	Source
gf	-524.68	kJ/mol	Joback Method
hf	-776.96	kJ/mol	Joback Method
hfus	29.54	kJ/mol	Joback Method
hvap	67.87	kJ/mol	Joback Method
log10ws	0.08		Crippen Method
logp	-0.208		Crippen Method
mcvol	119.380	ml/mol	McGowan Method
pc	3901.37	kPa	Joback Method
rinsol	1292.00		NIST Webbook
tb	610.92	K	Joback Method
tc	799.49	K	Joback Method
tf	391.83	K	Joback Method
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.53	J/molxK	610.92	Joback Method
cpg	358.90	J/molxK	768.06	Joback Method
cpg	351.37	J/molxK	736.63	Joback Method
cpg	343.38	J/molxK	705.21	Joback Method
cpg	334.92	J/molxK	673.78	Joback Method
cpg	325.97	J/molxK	642.35	Joback Method
cpg	365.99	J/molxK	799.49	Joback Method
dvisc	0.0002682	Paxs	610.92	Joback Method
dvisc	0.0003576	Paxs	574.40	Joback Method

dvisc	0.0004960	Paxs	537.89	Joback Method
dvisc	0.0007214	Paxs	501.38	Joback Method
dvisc	0.0011129	Paxs	464.86	Joback Method
dvisc	0.0018484	Paxs	428.34	Joback Method
dvisc	0.0033747	Paxs	391.83	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R249246&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R249246&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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