

# Diethylmalonic acid, pentadecyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C30H58O4/c1-8-11-12-13-14-15-16-17-18-19-20-21-22-23-33-27(31)30(9-2,10
InchiKey:	ARLBEAMYUXWXBI-UHFFFAOYSA-N
Formula:	C30H58O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	482.78

## Physical Properties

Property code	Value	Unit	Source
gf	-262.88	kJ/mol	Joback Method
hf	-1174.91	kJ/mol	Joback Method
hfus	60.68	kJ/mol	Joback Method
hvap	97.71	kJ/mol	Joback Method
log10ws	-9.38		Crippen Method
logp	9.043		Crippen Method
mcvol	448.440	ml/mol	McGowan Method
pc	637.69	kPa	Joback Method
rinsol	2926.00		NIST Webbook
tb	1031.48	K	Joback Method
tc	1277.17	K	Joback Method
tf	562.02	K	Joback Method
vc	1.736	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1604.56	J/molxK	1031.48	Joback Method
cpg	1627.56	J/molxK	1072.43	Joback Method
cpg	1648.77	J/molxK	1113.38	Joback Method
cpg	1668.34	J/molxK	1154.33	Joback Method
cpg	1686.43	J/molxK	1195.27	Joback Method
cpg	1703.17	J/molxK	1236.22	Joback Method
cpg	1718.73	J/molxK	1277.17	Joback Method
dvisc	0.0002043	Paxs	562.02	Joback Method
dvisc	0.0000786	Paxs	640.26	Joback Method

dvisc	0.0000373	Paxs	718.51	Joback Method
dvisc	0.0000204	Paxs	796.75	Joback Method
dvisc	0.0000125	Paxs	874.99	Joback Method
dvisc	0.0000083	Paxs	953.24	Joback Method
dvisc	0.0000058	Paxs	1031.48	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369487&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369487&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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