

# Butane, 2,3-dichloro-, (R\*,S\*)-

<b>Other names:</b>	Butane, 2,3-dichloro-, meso-meso-2,3-Dichlorobutane
<b>Inchi:</b>	InChI=1S/C4H8Cl2/c1-3(5)4(2)6/h3-4H,1-2H3/t3-,4+
<b>InchiKey:</b>	RMISVOPUIFJTEO-ZXZARUISSA-N
<b>Formula:</b>	C4H8Cl2
<b>SMILES:</b>	CC(Cl)C(C)Cl
<b>Mol. weight [g/mol]:</b>	127.01
<b>CAS:</b>	4028-56-2

## Physical Properties

Property code	Value	Unit	Source
chl	-2523.90 ± 1.70	kJ/mol	NIST Webbook
gf	-45.94	kJ/mol	Joback Method
hf	-202.30 ± 1.70	kJ/mol	NIST Webbook
hfl	-240.80 ± 1.70	kJ/mol	NIST Webbook
hfus	7.46	kJ/mol	Joback Method
hvap	38.41 ± 0.07	kJ/mol	NIST Webbook
hvap	38.40	kJ/mol	NIST Webbook
log10ws	-2.03		Crippen Method
logp	2.241		Crippen Method
mcvol	91.700	ml/mol	McGowan Method
pc	3628.97	kPa	Joback Method
tb	364.90	K	Joback Method
tc	555.91	K	Joback Method
tf	164.68	K	Joback Method
vc	0.345	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	141.51	J/mol×K	364.90	Joback Method
cpg	149.43	J/mol×K	396.73	Joback Method
cpg	157.00	J/mol×K	428.57	Joback Method
cpg	164.21	J/mol×K	460.40	Joback Method

cpg	171.09	J/mol×K	492.24	Joback Method
cpg	177.64	J/mol×K	524.07	Joback Method
cpg	183.87	J/mol×K	555.91	Joback Method
dvisc	0.0130687	Paxs	164.68	Joback Method
dvisc	0.0042176	Paxs	198.05	Joback Method
dvisc	0.0018860	Paxs	231.42	Joback Method
dvisc	0.0010331	Paxs	264.79	Joback Method
dvisc	0.0006475	Paxs	298.16	Joback Method
dvisc	0.0004458	Paxs	331.53	Joback Method
dvisc	0.0003287	Paxs	364.90	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.28363e+01
Coeff. B	-2.46981e+03
Coeff. C	-9.48930e+01
Temperature range (K), min.	291.71
Temperature range (K), max.	423.11

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4028562&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4028562&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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