

Benzene, (bromoethynyl)-

Other names:	(Bromoethynyl)benzene Bromophenylacetylene Phenylethynyl bromide 1-Bromo-2-phenylacetylene 1-Phenyl-2-bromoacetylene
Inchi:	InChI=1S/C8H5Br/c9-7-6-8-4-2-1-3-5-8/h1-5H
InchiKey:	BPVHWNVBBDHIQU-UHFFFAOYSA-N
Formula:	C8H5Br
SMILES:	BrC#Cc1ccccc1
Mol. weight [g/mol]:	181.03
CAS:	932-87-6

Physical Properties

Property code	Value	Unit	Source
gf	346.01	kJ/mol	Joback Method
hf	326.71	kJ/mol	Joback Method
hfus	18.92	kJ/mol	Joback Method
hvap	44.27	kJ/mol	Joback Method
ie	8.65	eV	NIST Webbook
log10ws	-3.10		Crippen Method
logp	2.390		Crippen Method
mcvol	108.720	ml/mol	McGowan Method
pc	4917.69	kPa	Joback Method
tb	484.28	K	Joback Method
tc	746.44	K	Joback Method
tf	372.24	K	Joback Method
vc	0.400	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	180.08	J/mol×K	484.28	Joback Method
cpg	190.59	J/mol×K	527.97	Joback Method
cpg	200.23	J/mol×K	571.67	Joback Method

cpg	209.06	J/mol×K	615.36	Joback Method
cpg	217.15	J/mol×K	659.05	Joback Method
cpg	224.55	J/mol×K	702.74	Joback Method
cpg	231.33	J/mol×K	746.44	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C932876&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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