

# Benzaldehyde, 3-hydroxy-4-methoxy, O-methylloxime

Inchi:	InChI=1S/C9H11NO3/c1-12-9-4-3-7(5-8(9)11)6-10-13-2/h3-6,11H,1-2H3/b10-6+
InchiKey:	LLQDRFKTWYEAKV-UXBLZVDNSA-N
Formula:	C9H11NO3
SMILES:	CON=Cc1ccc(OC)c(O)c1
Mol. weight [g/mol]:	181.19

## Physical Properties

Property code	Value	Unit	Source
hf	-363.56	kJ/mol	Joback Method
hvap	59.71	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	1.381		Crippen Method
mcvol	137.200	ml/mol	McGowan Method
pc	3243.03	kPa	Joback Method
rinpol	1623.00		NIST Webbook
tb	639.12	K	Joback Method
tc	872.83	K	Joback Method

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R100042&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R100042&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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