

# D-Leucine

Other names:	(R)-(-)-Leucine (S)-(+)-leucine (S)-2-amino-4-methylpentanoic acid (S)-2-amino-4-methylvaleric acid (S)-leucine .alpha.-amino-.gamma.-methylvaleric acid .alpha.-aminoisocaproic acid 4-methyl-L-norvaline L-(+)-leucine L.alpha.-aminoisocaproic acid L-leucine Leucine, D-
Inchi:	InChI=1S/C6H13NO2/c1-4(2)3-5(7)6(8)9/h4-5H,3,7H2,1-2H3,(H,8,9)/t5-/m0/s1
InchiKey:	ROHFNLRQFUQHCH-YFKPBYRVSA-N
Formula:	C6H13NO2
SMILES:	CC(C)CC(N)C(=O)O
Mol. weight [g/mol]:	131.17
CAS:	328-38-1

## Physical Properties

Property code	Value	Unit	Source
chs	-3581.40 ± 0.84	kJ/mol	NIST Webbook
gf	-204.53	kJ/mol	Joback Method
hf	-408.75	kJ/mol	Joback Method
hfs	-637.56 ± 0.84	kJ/mol	NIST Webbook
hfus	15.13	kJ/mol	Joback Method
hvap	62.24	kJ/mol	Joback Method
log10ws	-0.74		Crippen Method
logp	0.444		Crippen Method
mcvol	112.820	ml/mol	McGowan Method
pc	4077.71	kPa	Joback Method
tb	554.38	K	Joback Method
tc	743.19	K	Joback Method
tf	321.39	K	Joback Method
vc	0.413	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.71	J/molxK	554.38	Joback Method
cpg	285.39	J/molxK	585.85	Joback Method
cpg	294.59	J/molxK	617.32	Joback Method
cpg	303.34	J/molxK	648.79	Joback Method
cpg	311.64	J/molxK	680.26	Joback Method
cpg	319.50	J/molxK	711.72	Joback Method
cpg	326.94	J/molxK	743.19	Joback Method
psub	9.00e-04	kPa	420.70	Sublimation and vapour pressure estimation of l-leucine using thermogravimetric analysis
psub	2.90e-03	kPa	440.30	Sublimation and vapour pressure estimation of l-leucine using thermogravimetric analysis
psub	0.04	kPa	469.90	Sublimation and vapour pressure estimation of l-leucine using thermogravimetric analysis
psub	0.53	kPa	499.20	Sublimation and vapour pressure estimation of l-leucine using thermogravimetric analysis
psub	2.16	kPa	517.50	Sublimation and vapour pressure estimation of l-leucine using thermogravimetric analysis

## Sources

Intermolecular interactions of .alpha.-amino acids and glycyl  
interactions with the homologous series of  
promine acids with asparagine in presence  
interactions in valanine, l-proline  
at different temperatures and spectroscopy  
of valine, l-leucine aqueous KCl/KNO<sub>3</sub>  
systems at different temperatures. An  
amphoteric nature of l-leucine in aqueous  
alpha-amino acids in aqueous sodium  
fluoride solutions at different  
temperatures:

<https://www.doi.org/10.1016/j.jct.2016.06.018>  
<https://www.doi.org/10.1016/j.jct.2013.09.009>  
<https://www.doi.org/10.1016/j.tca.2008.10.023>  
<https://www.doi.org/10.1016/j.jct.2011.01.004>

<https://www.doi.org/10.1021/acs.jced.6b00168>

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>psub:</b>	Sublimation pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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