

# 1-Octene, 2-methyl-

<b>Other names:</b>	2-METHYLOCT-1-ENE 2-Methyl-1-octene
<b>Inchi:</b>	InChI=1S/C9H18/c1-4-5-6-7-8-9(2)3/h2,4-8H2,1,3H3
<b>InchiKey:</b>	FBEDQPGLIKZGIN-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	C=C(C)CCCCC
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	4588-18-5

## Physical Properties

Property code	Value	Unit	Source
gf	104.19	kJ/mol	Joback Method
hf	-113.45	kJ/mol	Joback Method
hfus	16.48	kJ/mol	Joback Method
hvap	35.04	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.533		Crippen Method
mcvol	133.370	ml/mol	McGowan Method
pc	2414.74	kPa	Joback Method
rinpol	874.00		NIST Webbook
rinpol	883.00		NIST Webbook
rinpol	874.00		NIST Webbook
rinpol	858.00		NIST Webbook
rinpol	882.00		NIST Webbook
rinpol	880.00		NIST Webbook
rinpol	898.00		NIST Webbook
rinpol	880.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	898.00		NIST Webbook
rinpol	881.00		NIST Webbook
rinpol	881.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	874.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	874.00		NIST Webbook
rinpol	876.00		NIST Webbook
rinpol	860.00		NIST Webbook

rinp	880.00		NIST Webbook
tb	415.00 ± 5.00	K	NIST Webbook
tb	416.00 ± 4.00	K	NIST Webbook
tb	416.30 ± 2.00	K	NIST Webbook
tb	418.00 ± 0.20	K	NIST Webbook
tb	417.61 ± 0.30	K	NIST Webbook
tb	417.95 ± 0.25	K	NIST Webbook
tc	571.36	K	Joback Method
tf	195.29 ± 0.20	K	NIST Webbook
vc	0.521	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.27	J/mol×K	401.88	Joback Method
cpg	271.11	J/mol×K	430.13	Joback Method
cpg	284.40	J/mol×K	458.37	Joback Method
cpg	297.17	J/mol×K	486.62	Joback Method
cpg	309.42	J/mol×K	514.87	Joback Method
cpg	321.16	J/mol×K	543.11	Joback Method
cpg	332.43	J/mol×K	571.36	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.44084e+01
Coeff. B	-3.55904e+03
Coeff. C	-5.34660e+01
Temperature range (K), min.	305.51
Temperature range (K), max.	444.70

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4588185&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4588185&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol342.mol">https://www.cheric.org/files/research/kdb/mol/mol342.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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