

# Ethyl 3,4-dihydroxybenzoate

|                      |                                                                                                      |
|----------------------|------------------------------------------------------------------------------------------------------|
| Other names:         | Ethyl protocatchuate<br>Protocatechuic acid ethyl ester<br>Benzoic acid, 3,4-dihydroxy-, ethyl ester |
| Inchi:               | InChI=1S/C9H10O4/c1-2-13-9(12)6-3-4-7(10)8(11)5-6/h3-5,10-11H,2H2,1H3                                |
| InchiKey:            | KBPUBCVJHFYPOC-UHFFFAOYSA-N                                                                          |
| Formula:             | C9H10O4                                                                                              |
| SMILES:              | CCOC(=O)c1ccc(O)c(O)c1                                                                               |
| Mol. weight [g/mol]: | 182.17                                                                                               |
| CAS:                 | 3943-89-3                                                                                            |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -405.85 | kJ/mol  | Joback Method  |
| hf            | -591.98 | kJ/mol  | Joback Method  |
| hfus          | 27.46   | kJ/mol  | Joback Method  |
| hvap          | 73.09   | kJ/mol  | Joback Method  |
| log10ws       | -1.23   |         | Crippen Method |
| logp          | 1.274   |         | Crippen Method |
| mcvol         | 133.090 | ml/mol  | McGowan Method |
| pc            | 5051.40 | kPa     | Joback Method  |
| tb            | 669.53  | K       | Joback Method  |
| tc            | 906.17  | K       | Joback Method  |
| tf            | 513.21  | K       | Joback Method  |
| vc            | 0.388   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 346.64 | J/mol×K | 669.53          | Joback Method |
| cpg           | 390.46 | J/mol×K | 866.73          | Joback Method |
| cpg           | 382.41 | J/mol×K | 827.29          | Joback Method |
| cpg           | 374.12 | J/mol×K | 787.85          | Joback Method |
| cpg           | 365.48 | J/mol×K | 748.41          | Joback Method |
| cpg           | 356.36 | J/mol×K | 708.97          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 398.39    | J/mol×K | 906.17 | Joback Method |
| dvisc | 0.0000027 | Pa×s    | 669.53 | Joback Method |
| dvisc | 0.0000041 | Pa×s    | 643.48 | Joback Method |
| dvisc | 0.0000062 | Pa×s    | 617.42 | Joback Method |
| dvisc | 0.0000100 | Pa×s    | 591.37 | Joback Method |
| dvisc | 0.0000167 | Pa×s    | 565.32 | Joback Method |
| dvisc | 0.0000293 | Pa×s    | 539.26 | Joback Method |
| dvisc | 0.0000545 | Pa×s    | 513.21 | Joback Method |

## Sources

|                        |                                                                                                                                             |
|------------------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3943893&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3943893&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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