

# 2,4,6-Trimethylpiperidine

<b>Inchi:</b>	InChI=1S/C8H17N/c1-6-4-7(2)9-8(3)5-6/h6-9H,4-5H2,1-3H3
<b>InchiKey:</b>	GELXVDNNZSIRQA-UHFFFAOYSA-N
<b>Formula:</b>	C8H17N
<b>SMILES:</b>	CC1CC(C)NC(C)C1
<b>Mol. weight [g/mol]:</b>	127.23
<b>CAS:</b>	21974-48-1

## Physical Properties

Property code	Value	Unit	Source
gf	113.22	kJ/mol	Joback Method
hf	-157.00	kJ/mol	Joback Method
hfus	20.04	kJ/mol	Joback Method
hvap	39.97	kJ/mol	Joback Method
log10ws	-2.24		Crippen Method
logp	1.783		Crippen Method
mcvol	122.700	ml/mol	McGowan Method
pc	3012.33	kPa	Joback Method
tb	441.20	K	Joback Method
tc	649.34	K	Joback Method
tf	283.85	K	Joback Method
vc	0.452	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.28	J/molxK	441.20	Joback Method
cpg	276.60	J/molxK	475.89	Joback Method
cpg	294.15	J/molxK	510.58	Joback Method
cpg	310.95	J/molxK	545.27	Joback Method
cpg	326.99	J/molxK	579.96	Joback Method
cpg	342.28	J/molxK	614.65	Joback Method
cpg	356.81	J/molxK	649.34	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.34660e+01
Coeff. B	-3.28964e+03
Coeff. C	-5.73960e+01
Temperature range (K), min.	272.15
Temperature range (K), max.	460.81

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21974481&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21974481&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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