

2-Oxepanone

Other names:	.epsilon.-caprolactone 1,6-Hexanolide 1-Oxa-2-oxocycloheptane 2-Oxacycloheptanone 6-Caprolactone 6-Caprolactone monomer 6-Hexanalactone 6-Hexanolactone 6-Hexanolide 6-Hydroxyhexan-6-olide 6-Hydroxyhexanoic acid lactone Caprolactone E-caprolactone Epsilon-caprolactone Hexamethylene oxide, 2-oxo- Hexan-6-olide Hexanoic acid, 6-hydroxy-, lactone Hexanoic acid, 6-hydroxy-, «epsilon»-lactone Hexanoic acid, 6-hydroxy-, «epsilon»-lactone Hexanoic acid, «epsilon»-lactone Hexanoic acid, «epsilon»-lactone Placel M oxepan-2-one «epsilon»-Caprolactone «epsilon»-Caprolactone monomer «epsilon»-Kapolakton «epsilon»-Caprolactone «epsilon»-Caprolactone monomer «epsilon»-Kapolakton
Inchi:	InChI=1S/C6H10O2/c7-6-4-2-1-3-5-8-6/h1-5H2
InchiKey:	PAPBSGBWRJIAAV-UHFFFAOYSA-N
Formula:	C6H10O2
SMILES:	O=C1CCCCO1
Mol. weight [g/mol]:	114.14
CAS:	502-44-3

Physical Properties

Property code	Value	Unit	Source
gf	-189.01	kJ/mol	Joback Method
hf	-396.00 ± 3.00	kJ/mol	NIST Webbook
hfl	-458.00 ± 2.00	kJ/mol	NIST Webbook
hfus	7.45	kJ/mol	Joback Method
hvap	62.00	kJ/mol	NIST Webbook
hvap	62.00 ± 1.00	kJ/mol	NIST Webbook
hvap	62.00 ± 1.30	kJ/mol	NIST Webbook
log10ws	-1.09		Crippen Method
logp	1.104		Crippen Method
mcvol	91.980	ml/mol	McGowan Method
pc	4414.96	kPa	Joback Method
sl	235.60	J/mol×K	NIST Webbook
sl	235.70	J/mol×K	NIST Webbook
tb	459.94	K	Joback Method
tc	695.47	K	Joback Method
tf	260.27	K	Joback Method
tt	272.13 ± 0.02	K	NIST Webbook
tt	271.83 ± 0.02	K	NIST Webbook
vc	0.326	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	256.71	J/mol×K	656.22	Joback Method
cpg	232.06	J/mol×K	577.71	Joback Method
cpg	218.67	J/mol×K	538.45	Joback Method
cpg	204.60	J/mol×K	499.20	Joback Method
cpg	189.87	J/mol×K	459.94	Joback Method
cpg	267.94	J/mol×K	695.47	Joback Method
cpg	244.74	J/mol×K	616.96	Joback Method
cpl	196.80	J/mol×K	298.15	NIST Webbook
cpl	196.80	J/mol×K	298.15	NIST Webbook
dvisc	0.0048852	Paxs	303.15	Volumetric and transport properties of binary liquid mixtures of N-methylacetamide with lactones at temperatures (303.15 to 318.15) K

dvisc	0.0043237	Paxs	308.15	Volumetric and transport properties of binary liquid mixtures of N-methylacetamide with lactones at temperatures (303.15 to 318.15) K
dvisc	0.0038182	Paxs	313.15	Volumetric and transport properties of binary liquid mixtures of N-methylacetamide with lactones at temperatures (303.15 to 318.15) K
dvisc	0.0034203	Paxs	318.15	Volumetric and transport properties of binary liquid mixtures of N-methylacetamide with lactones at temperatures (303.15 to 318.15) K
hfust	13.82	kJ/mol	272.00	NIST Webbook
hfust	13.82	kJ/mol	271.83	NIST Webbook
hfust	13.82	kJ/mol	272.13	NIST Webbook
hvapt	54.00 ± 0.20	kJ/mol	415.50	NIST Webbook
pvap	0.04	kPa	283.20	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
pvap	0.34	kPa	323.10	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
pvap	0.21	kPa	313.10	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
pvap	0.13	kPa	302.90	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone

pvap	0.08	kPa	292.90	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
pvap	0.52	kPa	333.20	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
pvap	0.77	kPa	343.20	Vapour pressure data of e-caprolactone, d-hexalactone, and c-caprolactone
sfust	50.90	J/molxK	271.83	NIST Webbook
sfust	50.90	J/molxK	272.13	NIST Webbook
speedsl	1567.76	m/s	293.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones
speedsl	1548.28	m/s	298.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones
speedsl	1528.99	m/s	303.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones
speedsl	1509.98	m/s	308.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones
speedsl	1491.20	m/s	313.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones

speedsl	1472.68	m/s	318.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones
speedsl	1454.52	m/s	323.15	A comparative study on the interactions of [bmim][NTf2] ionic liquid with selected four- to seven-membered-ring lactones

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	370.20	K	2.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	3.47420e+01
Coeff. B	-1.27202e+04
Coeff. C	-4.53800e+00
Temperature range (K), min.	373.73
Temperature range (K), max.	436.75

Sources

Measurement and Correlation for Solubilities of Adipic Acid in Acetic Acid, Propylene Carbonate, and High Pressure Phase Equilibria Data for the Ternary System Containing Dichloromethane, and epsilon-Caprolactone: Vapour pressure data of e-caprolactone, d-hexalactone, and epsilon-caprolactone.

<https://www.doi.org/10.1021/acs.jced.6b00177>

<https://www.doi.org/10.1021/acs.jced.8b01017>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

<https://www.doi.org/10.1016/j.jct.2007.09.008>

<http://link.springer.com/article/10.1007/BF02311772>

The Yaws Handbook of Vapor Pressure:
NIST Webbook:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Phase behaviour of binary systems of lactones in carbon dioxide: A comparative study on the interactions of [bmim][NTf₂] ionic liquid with ethylene carbonate
Measurement and Correlation for Solubilities of Succinic Acid and Gluconic Acid in ε-Caprolactone + Acetic Acid Mixtures and ε-Caprolactone + Cyclohexanone Binary Liquid mixtures of N-Methylacetamide with lactones at temperatures (303.15 to 318.15) K:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C502443&Units=SI>

<https://www.doi.org/10.1016/j.jct.2009.07.003>

<https://www.doi.org/10.1016/j.jct.2016.12.032>

<https://www.doi.org/10.1021/acs.jced.7b00660>

https://www.chemeo.com/doc/models/crippen_log10ws

<https://www.doi.org/10.1016/j.jct.2008.05.004>

https://en.wikipedia.org/wiki/Joback_method

Legend

cp_g:	Ideal gas heat capacity
cp_l:	Liquid phase heat capacity
d_{visc}:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
h_{fl}:	Liquid phase enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{fust}:	Enthalpy of fusion at a given temperature
h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log_{10ws}:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
pv_{ap}:	Vapor pressure
sf_{ust}:	Entropy of fusion at a given temperature
sl:	Liquid phase molar entropy at standard conditions
speed_{sl}:	Speed of sound in fluid
tb:	Normal Boiling Point Temperature
t_{brp}:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature
vc:	Critical Volume

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