

# 2-Hydroxyethyl 2-(2,4,5-trichlorophenoxy)propanoate

Inchi:	InChI=1S/C11H11Cl3O4/c1-6(11(16)17-3-2-15)18-10-5-8(13)7(12)4-9(10)14/h4-6,15H,2-
InchiKey:	PDVSFMMXQVCJGE-UHFFFAOYSA-N
Formula:	C11H11Cl3O4
SMILES:	CC(Oc1cc(Cl)c(Cl)cc1Cl)C(=O)OCCO
Mol. weight [g/mol]:	313.56
CAS:	116402-64-3

## Physical Properties

Property code	Value	Unit	Source
gf	-388.71	kJ/mol	Joback Method
hf	-650.00	kJ/mol	Joback Method
hfus	34.25	kJ/mol	Joback Method
hvap	85.35	kJ/mol	Joback Method
log10ws	-3.56		Crippen Method
logp	2.950		Crippen Method
mcvol	197.990	ml/mol	McGowan Method
pc	2576.72	kPa	Joback Method
tb	795.44	K	Joback Method
tc	1006.75	K	Joback Method
tf	507.68	K	Joback Method
vc	0.746	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	487.50	J/molxK	795.44	Joback Method
cpg	524.88	J/molxK	971.53	Joback Method
cpg	518.85	J/molxK	936.31	Joback Method
cpg	512.09	J/molxK	901.10	Joback Method
cpg	504.61	J/molxK	865.88	Joback Method
cpg	496.41	J/molxK	830.66	Joback Method
cpg	530.18	J/molxK	1006.75	Joback Method
dvisc	0.0000215	Paxs	795.44	Joback Method
dvisc	0.0000296	Paxs	747.48	Joback Method

dvisc	0.0000427	Paxs	699.52	Joback Method
dvisc	0.0000650	Paxs	651.56	Joback Method
dvisc	0.0001057	Paxs	603.60	Joback Method
dvisc	0.0001870	Paxs	555.64	Joback Method
dvisc	0.0003683	Paxs	507.68	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402643&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402643&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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