

# Fumaric acid, dodecyl 4-phenylphenyl ester

<b>Inchi:</b>	InChI=1S/C28H36O4/c1-2-3-4-5-6-7-8-9-10-14-23-31-27(29)21-22-28(30)32-26-19-17-25
<b>InchiKey:</b>	VKJFZZMMKRBKKM-QURGRASLSA-N
<b>Formula:</b>	C28H36O4
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)C=CC(=O)Oc1ccc(-c2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	436.58

## Physical Properties

Property code	Value	Unit	Source
gf	12.45	kJ/mol	Joback Method
hf	-532.04	kJ/mol	Joback Method
hfus	61.74	kJ/mol	Joback Method
hvap	101.41	kJ/mol	Joback Method
log10ws	-8.96		Crippen Method
logp	7.279		Crippen Method
mvol	368.440	ml/mol	McGowan Method
pc	1030.59	kPa	Joback Method
rinpol	3616.00		NIST Webbook
rinpol	3616.00		NIST Webbook
tb	1055.12	K	Joback Method
tc	1291.87	K	Joback Method
tf	609.92	K	Joback Method
vc	1.415	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1231.48	J/molxK	1055.12	Joback Method
cpg	1293.09	J/molxK	1252.42	Joback Method
cpg	1283.11	J/molxK	1212.96	Joback Method
cpg	1272.06	J/molxK	1173.50	Joback Method
cpg	1259.84	J/molxK	1134.04	Joback Method
cpg	1246.35	J/molxK	1094.58	Joback Method
cpg	1302.10	J/molxK	1291.87	Joback Method
dvisc	0.0000153	Paxs	1055.12	Joback Method

dvisc	0.0000200	Paxs	980.92	Joback Method
dvisc	0.0000271	Paxs	906.72	Joback Method
dvisc	0.0000389	Paxs	832.52	Joback Method
dvisc	0.0000599	Paxs	758.32	Joback Method
dvisc	0.0001014	Paxs	684.12	Joback Method
dvisc	0.0001949	Paxs	609.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348218&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348218&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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