

# 2,4A-Dihydro-fluorene

<b>Inchi:</b>	InChI=1S/C13H12/c1-3-7-12-10(5-1)9-11-6-2-4-8-13(11)12/h1-3,5-8,11H,4,9H2
<b>InchiKey:</b>	WRUSSAHPTVRSHH-UHFFFAOYSA-N
<b>Formula:</b>	C13H12
<b>SMILES:</b>	C1=CC2Cc3ccccc3C2=CC1
<b>Mol. weight [g/mol]:</b>	168.23

## Physical Properties

Property code	Value	Unit	Source
gf	328.76	kJ/mol	Joback Method
hf	177.28	kJ/mol	Joback Method
hfus	18.23	kJ/mol	Joback Method
hvap	49.02	kJ/mol	Joback Method
log10ws	-3.75		Crippen Method
logp	3.202		Crippen Method
mcvol	139.950	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
rinpol	258.60		NIST Webbook
rinpol	258.60		NIST Webbook
tb	554.22	K	Joback Method
tc	798.24	K	Joback Method
tf	325.85	K	Joback Method
vc	0.534	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	327.93	J/molxK	554.22	Joback Method
cpg	344.80	J/molxK	594.89	Joback Method
cpg	360.27	J/molxK	635.56	Joback Method
cpg	374.45	J/molxK	676.23	Joback Method
cpg	387.50	J/molxK	716.90	Joback Method
cpg	399.53	J/molxK	757.57	Joback Method
cpg	410.67	J/molxK	798.24	Joback Method
dvisc	0.0015943	Paxs	325.85	Joback Method

dvisc	0.0012721	Paxs	363.91	Joback Method
dvisc	0.0010594	Paxs	401.97	Joback Method
dvisc	0.0009106	Paxs	440.04	Joback Method
dvisc	0.0008018	Paxs	478.10	Joback Method
dvisc	0.0007194	Paxs	516.16	Joback Method
dvisc	0.0006551	Paxs	554.22	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R508659&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R508659&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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