

# 1H-3a,7-Methanoazulen-5-ol, octahydro-3,8,8-trimethyl-6-methylene-

Other names:

Cedr-8(15)-en-9-ol

Cedrenol

«beta»-cedren-9-«alpha»-ol

octahydro-3,8,8-trimethyl-6-methylene-1H-3a,7-methanoazulen-5-ol

Inchi:

InChI=1S/C15H24O/c1-9-5-6-13-14(3,4)11-7-15(9,13)8-12(16)10(11)2/h9,11-13,16H,2,5

InchiKey:

DJYWGTBEZVORGE-UHFFFAOYSA-N

Formula:

C15H24O

SMILES:

C=C1C(O)CC23CC1C(C)(C)C2CCC3C

Mol. weight [g/mol]:

220.35

CAS:

28231-03-0

## Physical Properties

Property code	Value	Unit	Source
gf	115.62	kJ/mol	Joback Method
hf	-245.38	kJ/mol	Joback Method
hfus	18.36	kJ/mol	Joback Method
hvap	62.68	kJ/mol	Joback Method
log10ws	-3.81		Crippen Method
logp	3.386		Crippen Method
mcvol	191.200	ml/mol	McGowan Method
pc	2233.41	kPa	Joback Method
rinpol	1615.00		NIST Webbook
rinpol	1604.00		NIST Webbook
rinpol	1645.00		NIST Webbook
rinpol	1615.00		NIST Webbook
rinpol	1645.00		NIST Webbook
rinpol	1604.00		NIST Webbook
ripol	2110.00		NIST Webbook
ripol	2129.00		NIST Webbook
ripol	2133.00		NIST Webbook
ripol	2133.00		NIST Webbook
ripol	2110.00		NIST Webbook
ripol	2113.00		NIST Webbook
ripol	2142.00		NIST Webbook
ripol	2115.00		NIST Webbook
ripol	2129.00		NIST Webbook
tb	649.17	K	Joback Method

tc	856.86	K	Joback Method
tf	415.17	K	Joback Method
vc	0.727	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	575.53	J/mol×K	649.17	Joback Method
cpg	594.60	J/mol×K	683.79	Joback Method
cpg	612.82	J/mol×K	718.40	Joback Method
cpg	630.40	J/mol×K	753.02	Joback Method
cpg	647.59	J/mol×K	787.63	Joback Method
cpg	664.58	J/mol×K	822.25	Joback Method
cpg	681.62	J/mol×K	856.86	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C28231030&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C28231030&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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