

# Benzoylacrylic acid

<b>Other names:</b>	2-Butenoic acid, 4-oxo-4-phenyl- 3-Benzoylacrylic acid Acrylic acid, 3-benzoyl- Crotonic acid, 4-oxo-4-phenyl- «beta»-Benzoylacrylic acid 4-Oxo-4-phenyl-2-butenoic acid NSC 143
<b>Inchi:</b>	InChI=1S/C10H8O3/c11-9(6-7-10(12)13)8-4-2-1-3-5-8/h1-7H,(H,12,13)/b7-6+
<b>InchiKey:</b>	PLPDHGOODMBBGN-VOTSOKGWSA-N
<b>Formula:</b>	C10H8O3
<b>SMILES:</b>	O=C(O)C=CC(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	176.17
<b>CAS:</b>	583-06-2

## Physical Properties

Property code	Value	Unit	Source
gf	-168.71	kJ/mol	Joback Method
hf	-273.37	kJ/mol	Joback Method
hfus	23.18	kJ/mol	Joback Method
hvap	70.26	kJ/mol	Joback Method
log10ws	-1.92		Crippen Method
logp	1.510		Crippen Method
mcvol	132.710	ml/mol	McGowan Method
pc	4021.02	kPa	Joback Method
tb	658.96	K	Joback Method
tc	874.35	K	Joback Method
tf	384.48	K	Joback Method
vc	0.498	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	317.48	J/mol×K	658.96	Joback Method
cpg	326.80	J/mol×K	694.86	Joback Method

cpg	335.42	J/molxK	730.76	Joback Method
cpg	343.40	J/molxK	766.66	Joback Method
cpg	350.78	J/molxK	802.56	Joback Method
cpg	357.61	J/molxK	838.45	Joback Method
cpg	363.94	J/molxK	874.35	Joback Method
dvisc	0.0029239	Paxs	384.48	Joback Method
dvisc	0.0011188	Paxs	430.23	Joback Method
dvisc	0.0005149	Paxs	475.97	Joback Method
dvisc	0.0002715	Paxs	521.72	Joback Method
dvisc	0.0001587	Paxs	567.47	Joback Method
dvisc	0.0001006	Paxs	613.21	Joback Method
dvisc	0.0000679	Paxs	658.96	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C583062&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C583062&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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