

# Tetrahydrofuran, 3-chloro-2-[2-chloro-1-(chloromethyl)ethoxy]

**Other names:** 2-(2-Chloro-1-chloromethyl-ethoxy)- 3-chloro-tetrahydro-furan  
**Inchi:** InChI=1S/C7H11Cl3O2/c8-3-5(4-9)12-7-6(10)1-2-11-7/h5-7H,1-4H2  
**InchiKey:** QFPRNAFYFZCLKW-UHFFFAOYSA-N  
**Formula:** C7H11Cl3O2  
**SMILES:** CICC(CCl)OC1OCCC1Cl  
**Mol. weight [g/mol]:** 233.52

## Physical Properties

Property code	Value	Unit	Source
gf	-192.45	kJ/mol	Joback Method
hf	-464.39	kJ/mol	Joback Method
hfus	27.13	kJ/mol	Joback Method
hvap	50.81	kJ/mol	Joback Method
log10ws	-2.11		Crippen Method
logp	2.203		Crippen Method
mcvol	147.090	ml/mol	McGowan Method
pc	2853.57	kPa	Joback Method
rinpol	1435.00		NIST Webbook
rinpol	1435.00		NIST Webbook
rinpol	1435.00		NIST Webbook
tb	531.39	K	Joback Method
tc	747.74	K	Joback Method
tf	298.87	K	Joback Method
vc	0.547	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	314.79	J/molxK	531.39	Joback Method
cpg	328.18	J/molxK	567.45	Joback Method
cpg	340.81	J/molxK	603.51	Joback Method
cpg	352.70	J/molxK	639.57	Joback Method
cpg	363.87	J/molxK	675.62	Joback Method
cpg	374.32	J/molxK	711.68	Joback Method

cpg	384.07	J/molxK	747.74	Joback Method
dvisc	0.0033336	Paxs	298.87	Joback Method
dvisc	0.0018499	Paxs	337.62	Joback Method
dvisc	0.0011589	Paxs	376.38	Joback Method
dvisc	0.0007922	Paxs	415.13	Joback Method
dvisc	0.0005779	Paxs	453.88	Joback Method
dvisc	0.0004430	Paxs	492.64	Joback Method
dvisc	0.0003531	Paxs	531.39	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R91208&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R91208&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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