

# Formic acid, 3-methylbut-2-yl ester

Inchi:	InChI=1S/C6H12O2/c1-5(2)6(3)8-4-7/h4-6H,1-3H3
InchiKey:	XEQSPMCNFJQDHU-UHFFFAOYSA-N
Formula:	C6H12O2
SMILES:	CC(C)C(C)OC=O
Mol. weight [g/mol]:	116.16

## Physical Properties

Property code	Value	Unit	Source
gf	-209.76	kJ/mol	Joback Method
hf	-395.53	kJ/mol	Joback Method
hfus	7.73	kJ/mol	Joback Method
hvap	37.30	kJ/mol	Joback Method
log10ws	-1.07		Crippen Method
logp	1.204		Crippen Method
mcvol	102.840	ml/mol	McGowan Method
pc	3376.28	kPa	Joback Method
rinpol	763.00		NIST Webbook
tb	406.88	K	Joback Method
tc	587.63	K	Joback Method
tf	191.61	K	Joback Method
vc	0.395	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	201.71	J/molxK	406.88	Joback Method
cpg	211.86	J/molxK	437.00	Joback Method
cpg	221.67	J/molxK	467.13	Joback Method
cpg	231.14	J/molxK	497.25	Joback Method
cpg	240.27	J/molxK	527.38	Joback Method
cpg	249.06	J/molxK	557.50	Joback Method
cpg	257.51	J/molxK	587.63	Joback Method
dvisc	0.0088875	Paxs	191.61	Joback Method
dvisc	0.0031355	Paxs	227.49	Joback Method

dvisc	0.0014693	Paxs	263.37	Joback Method
dvisc	0.0008258	Paxs	299.25	Joback Method
dvisc	0.0005250	Paxs	335.12	Joback Method
dvisc	0.0003644	Paxs	371.00	Joback Method
dvisc	0.0002697	Paxs	406.88	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U367952&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U367952&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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