

# Ethane, 1-ethoxy-2-methoxy-

<b>Other names:</b>	1-ethoxy-2-methoxyethane
<b>Inchi:</b>	InChI=1S/C5H12O2/c1-3-7-5-4-6-2/h3-5H2,1-2H3
<b>InchiKey:</b>	CAQYAZNFWDDMIT-UHFFFAOYSA-N
<b>Formula:</b>	C5H12O2
<b>SMILES:</b>	CCOCCOC
<b>Mol. weight [g/mol]:</b>	104.15
<b>CAS:</b>	5137-45-1

## Physical Properties

Property code	Value	Unit	Source
gf	-218.78	kJ/mol	Joback Method
hf	-410.97	kJ/mol	Joback Method
hfus	11.08	kJ/mol	Joback Method
hvap	39.83 ± 0.04	kJ/mol	NIST Webbook
hvap	39.80 ± 0.10	kJ/mol	NIST Webbook
log10ws	-0.09		Crippen Method
logp	0.669		Crippen Method
mcvol	93.050	ml/mol	McGowan Method
pc	3333.53	kPa	Joback Method
rinpol	680.50		NIST Webbook
rinpol	686.00		NIST Webbook
tb	358.00 ± 4.00	K	NIST Webbook
tb	376.65 ± 0.40	K	NIST Webbook
tb	376.00 ± 8.00	K	NIST Webbook
tc	524.09	K	Joback Method
tf	190.57	K	Joback Method
vc	0.351	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	169.54	J/mol×K	358.64	Joback Method
cpg	210.37	J/mol×K	496.51	Joback Method
cpg	202.53	J/mol×K	468.94	Joback Method

cpg	194.52	J/molxK	441.36	Joback Method
cpg	186.35	J/molxK	413.79	Joback Method
cpg	178.02	J/molxK	386.21	Joback Method
cpg	218.03	J/molxK	524.09	Joback Method
dvisc	0.0002027	Paxs	358.64	Joback Method
dvisc	0.0002582	Paxs	330.63	Joback Method
dvisc	0.0003441	Paxs	302.62	Joback Method
dvisc	0.0004861	Paxs	274.61	Joback Method
dvisc	0.0007430	Paxs	246.59	Joback Method
dvisc	0.0012659	Paxs	218.58	Joback Method
dvisc	0.0025228	Paxs	190.57	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5137451&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5137451&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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