

I-Valine, n-pentafluoropropionyl-, octadecyl ester

Inchi: InChI=1S/C26H46F5NO3/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-35-23(33)2
InchiKey: IVCBSNVFTOZHSI-UHFFFAOYSA-N
Formula: C26H46F5NO3
SMILES: CCCCCCCCCCCCCCCCCCOC(=O)C(NC(=O)C(F)(F)C(F)(F)F)C(C)C
Mol. weight [g/mol]: 515.64

Physical Properties

Property code	Value	Unit	Source
gf	-1078.66	kJ/mol	Joback Method
hf	-1892.49	kJ/mol	Joback Method
hfus	66.11	kJ/mol	Joback Method
hvap	88.35	kJ/mol	Joback Method
log10ws	-9.38		Crippen Method
logp	8.130		Crippen Method
mcvol	405.040	ml/mol	McGowan Method
pc	706.21	kPa	Joback Method
rinpol	2620.00		NIST Webbook
tb	963.62	K	Joback Method
tc	1195.91	K	Joback Method
tf	535.32	K	Joback Method
vc	1.613	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1413.61	J/mol×K	963.62	Joback Method
cpg	1434.72	J/mol×K	1002.34	Joback Method
cpg	1454.32	J/mol×K	1041.05	Joback Method
cpg	1472.57	J/mol×K	1079.77	Joback Method
cpg	1489.60	J/mol×K	1118.48	Joback Method
cpg	1505.56	J/mol×K	1157.20	Joback Method
cpg	1520.59	J/mol×K	1195.91	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U320892&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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