

# 2-Butenoic acid, 3-(methylamino)-, ethyl ester

<b>Other names:</b>	Crotonic acid, 3-(methylamino)-, ethyl ester Ethyl «beta»-(methylamino)crotonate Ethyl 3-(methylamino)-2-butenoate Ethyl 3-(methylamino)crotonate
<b>Inchi:</b>	InChI=1S/C7H13NO2/c1-4-10-7(9)5-6(2)8-3/h5,8H,4H2,1-3H3/b6-5+
<b>InchiKey:</b>	FARYTWBWLZAXNK-AATRIKPKSA-N
<b>Formula:</b>	C7H13NO2
<b>SMILES:</b>	CCOC(=O)C=C(C)NC
<b>Mol. weight [g/mol]:</b>	143.18
<b>CAS:</b>	870-85-9

## Physical Properties

Property code	Value	Unit	Source
gf	-64.80	kJ/mol	Joback Method
hf	-271.71	kJ/mol	Joback Method
hfus	20.66	kJ/mol	Joback Method
hvap	46.81	kJ/mol	Joback Method
log10ws	-1.15		Crippen Method
logp	0.673		Crippen Method
mcvol	122.610	ml/mol	McGowan Method
pc	3177.55	kPa	Joback Method
tb	490.06	K	Joback Method
tc	681.38	K	Joback Method
tf	274.43	K	Joback Method
vc	0.468	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	265.54	J/molxK	490.06	Joback Method
cpg	277.01	J/molxK	521.95	Joback Method
cpg	287.96	J/molxK	553.83	Joback Method
cpg	298.40	J/molxK	585.72	Joback Method
cpg	308.35	J/molxK	617.61	Joback Method

cpg	317.82	J/mol×K	649.50	Joback Method
cpg	326.82	J/mol×K	681.38	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C870859&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C870859&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/39-757-9/2-Butenoic-acid-3-methylamino-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-27 21:12:52.083012724 +0000 UTC m=+16541621.003590036.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.