

# 3-Phenoxypropyl acetate

<b>Inchi:</b>	InChI=1S/C11H14O3/c1-10(12)13-8-5-9-14-11-6-3-2-4-7-11/h2-4,6-7H,5,8-9H2,1H3
<b>InchiKey:</b>	FZOXUFODEISZLN-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O3
<b>SMILES:</b>	CC(=O)OCCCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	194.23

## Physical Properties

Property code	Value	Unit	Source
gf	-184.77	kJ/mol	Joback Method
hf	-410.86	kJ/mol	Joback Method
hfus	22.26	kJ/mol	Joback Method
hvap	53.92	kJ/mol	Joback Method
log10ws	-2.13		Crippen Method
logp	2.019		Crippen Method
mvol	155.400	ml/mol	McGowan Method
pc	2729.71	kPa	Joback Method
rinpol	1502.00		NIST Webbook
rinpol	1502.00		NIST Webbook
tb	576.47	K	Joback Method
tc	783.57	K	Joback Method
tf	334.54	K	Joback Method
vc	0.586	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.22	J/molxK	576.47	Joback Method
cpg	385.38	J/molxK	610.99	Joback Method
cpg	398.78	J/molxK	645.50	Joback Method
cpg	411.43	J/molxK	680.02	Joback Method
cpg	423.35	J/molxK	714.54	Joback Method
cpg	434.52	J/molxK	749.05	Joback Method
cpg	444.97	J/molxK	783.57	Joback Method
dvisc	0.0016964	Paxs	334.54	Joback Method

dvisc	0.0009299	Paxs	374.86	Joback Method
dvisc	0.0005728	Paxs	415.18	Joback Method
dvisc	0.0003845	Paxs	455.50	Joback Method
dvisc	0.0002753	Paxs	495.83	Joback Method
dvisc	0.0002073	Paxs	536.15	Joback Method
dvisc	0.0001625	Paxs	576.47	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378335&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378335&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/39-780-3/3-Phenoxypropyl-acetate.pdf>

Generated by Cheméo on 2024-04-19 22:25:18.913112866 +0000 UTC m=+15854767.833690182.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.