

# «beta»-Acoradienol

<b>Inchi:</b>	InChI=1S/C15H24O/c1-11(2)14-5-4-12(3)15(14)8-6-13(10-16)7-9-15/h6,12,14,16H,1,4-5
<b>InchiKey:</b>	KBMDEJULGPFFGC-UHFFFAOYSA-N
<b>Formula:</b>	C15H24O
<b>SMILES:</b>	<chem>C=C(C)C1CCC(C)C12CC=C(CO)CC2</chem>
<b>Mol. weight [g/mol]:</b>	220.35
<b>CAS:</b>	149496-35-5

## Physical Properties

Property code	Value	Unit	Source
gf	98.12	kJ/mol	Joback Method
hf	-227.35	kJ/mol	Joback Method
hfus	19.58	kJ/mol	Joback Method
hvap	65.08	kJ/mol	Joback Method
log10ws	-4.14		Crippen Method
logp	3.698		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	2212.45	kPa	Joback Method
rinpol	1736.00		NIST Webbook
rinpol	1732.00		NIST Webbook
rinpol	1732.00		NIST Webbook
rinpol	1732.00		NIST Webbook
rinpol	1732.00		NIST Webbook
rinpol	1736.00		NIST Webbook
rinpol	1754.00		NIST Webbook
rinpol	1764.10		NIST Webbook
rinpol	1768.00		NIST Webbook
rinpol	1741.00		NIST Webbook
rinpol	1757.00		NIST Webbook
tb	661.61	K	Joback Method
tc	869.92	K	Joback Method
tf	358.65	K	Joback Method
vc	0.742	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	572.16	J/mol×K	661.61	Joback Method
cpg	591.06	J/mol×K	696.33	Joback Method
cpg	608.99	J/mol×K	731.05	Joback Method
cpg	626.07	J/mol×K	765.76	Joback Method
cpg	642.44	J/mol×K	800.48	Joback Method
cpg	658.24	J/mol×K	835.20	Joback Method
cpg	673.58	J/mol×K	869.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C149496355&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C149496355&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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