

# Terephthalic acid, 2-chloropropyl decyl ester

**Inchi:** InChI=1S/C21H31ClO4/c1-3-4-5-6-7-8-9-10-15-25-20(23)18-11-13-19(14-12-18)21(24)20  
**InchiKey:** AUHPJSKSQHHXHG-UHFFFAOYSA-N  
**Formula:** C21H31ClO4  
**SMILES:** CCCCCCCCCCOC(=O)c1ccc(C(=O)OCC(C)Cl)cc1  
**Mol. weight [g/mol]:** 382.92

## Physical Properties

Property code	Value	Unit	Source
gf	-253.49	kJ/mol	Joback Method
hf	-762.33	kJ/mol	Joback Method
hfus	50.05	kJ/mol	Joback Method
hvap	87.59	kJ/mol	Joback Method
log10ws	-6.83		Crippen Method
logp	5.768		Crippen Method
mvol	310.110	ml/mol	McGowan Method
pc	1226.84	kPa	Joback Method
rinpol	2765.00		NIST Webbook
rinpol	2765.00		NIST Webbook
tb	901.11	K	Joback Method
tc	1109.38	K	Joback Method
tf	524.61	K	Joback Method
vc	1.194	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	964.82	J/molxK	901.11	Joback Method
cpg	1029.03	J/molxK	1074.67	Joback Method
cpg	1018.52	J/molxK	1039.96	Joback Method
cpg	1006.88	J/molxK	1005.25	Joback Method
cpg	994.07	J/molxK	970.53	Joback Method
cpg	980.06	J/molxK	935.82	Joback Method
cpg	1038.42	J/molxK	1109.38	Joback Method
dvisc	0.0000373	Paxs	901.11	Joback Method

dvisc	0.0000487	Paxs	838.36	Joback Method
dvisc	0.0000664	Paxs	775.61	Joback Method
dvisc	0.0000957	Paxs	712.86	Joback Method
dvisc	0.0001479	Paxs	650.11	Joback Method
dvisc	0.0002509	Paxs	587.36	Joback Method
dvisc	0.0004830	Paxs	524.61	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356177&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356177&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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