

3-Sulfanylhexyl Hexadodecanoate

Inchi: InChI=1S/C22H44O2S/c1-3-5-6-7-8-9-10-11-12-13-14-15-16-18-22(23)24-20-19-21(25)1
InchiKey: STJAWXNXWLRHTH-UHFFFAOYSA-N
Formula: C22H44O2S
SMILES: CCCCCCCCCCCCCCCC(=O)OCCC(S)CCC
Mol. weight [g/mol]: 372.65

Physical Properties

Property code	Value	Unit	Source
gf	-72.61	kJ/mol	Joback Method
hf	-709.01	kJ/mol	Joback Method
hfus	56.04	kJ/mol	Joback Method
hvap	80.07	kJ/mol	Joback Method
log10ws	-8.08		Crippen Method
logp	7.500		Crippen Method
mcvol	344.630	ml/mol	McGowan Method
pc	969.28	kPa	Joback Method
rinpol	2615.00		NIST Webbook
tb	841.47	K	Joback Method
tc	1032.20	K	Joback Method
tf	431.32	K	Joback Method
vc	1.339	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1099.51	J/mol×K	841.47	Joback Method
cpg	1119.42	J/mol×K	873.26	Joback Method
cpg	1138.17	J/mol×K	905.05	Joback Method
cpg	1155.82	J/mol×K	936.83	Joback Method
cpg	1172.40	J/mol×K	968.62	Joback Method
cpg	1187.95	J/mol×K	1000.41	Joback Method
cpg	1202.50	J/mol×K	1032.20	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R519669&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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