

# Phenol, 2-methyl-3,5-dinitro-

<b>Other names:</b>	o-Cresol, 3,5-dinitro- 3,5-Dinitro-o-cresol 4,6-Dinitro-2-hydroxytoluene 3,5-Dinitro-ortho-cresol
<b>Inchi:</b>	InChI=1S/C7H6N2O5/c1-4-6(9(13)14)2-5(8(11)12)3-7(4)10/h2-3,10H,1H3
<b>InchiKey:</b>	KSHJAFFDLKPUMT-UHFFFAOYSA-N
<b>Formula:</b>	C7H6N2O5
<b>SMILES:</b>	<chem>Cc1c(O)cc([N+](=O)[O-])cc1[N+](=O)[O-]</chem>
<b>Mol. weight [g/mol]:</b>	198.13
<b>CAS:</b>	497-56-3

## Physical Properties

Property code	Value	Unit	Source
gf	17.69	kJ/mol	Joback Method
hf	-173.05	kJ/mol	Joback Method
hfus	35.65	kJ/mol	Joback Method
hvap	80.97	kJ/mol	Joback Method
log10ws	-2.80		Crippen Method
logp	1.517		Crippen Method
mcvol	126.440	ml/mol	McGowan Method
pc	4987.39	kPa	Joback Method
tb	780.50	K	Joback Method
tc	1059.76	K	Joback Method
tf	619.05	K	Joback Method
vc	0.450	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.95	J/molxK	780.50	Joback Method
cpg	343.28	J/molxK	827.04	Joback Method
cpg	351.09	J/molxK	873.59	Joback Method
cpg	358.52	J/molxK	920.13	Joback Method
cpg	365.69	J/molxK	966.67	Joback Method

cpg	372.74	J/mol×K	1013.21	Joback Method
cpg	379.80	J/mol×K	1059.76	Joback Method
hsubt	103.30	kJ/mol	307.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C497563&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C497563&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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