

# Cyclobutanecarboxylic acid, 2-isopropoxyphenyl ester

Inchi:	InChI=1S/C14H18O3/c1-10(2)16-12-8-3-4-9-13(12)17-14(15)11-6-5-7-11/h3-4,8-11H,5-7
InchiKey:	LGYNTCSGORWHHN-UHFFFAOYSA-N
Formula:	C14H18O3
SMILES:	CC(C)Oc1ccccc1OC(=O)C1CCC1
Mol. weight [g/mol]:	234.29

## Physical Properties

Property code	Value	Unit	Source
gf	-122.93	kJ/mol	Joback Method
hf	-422.89	kJ/mol	Joback Method
hfus	22.16	kJ/mol	Joback Method
hvap	60.96	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	3.179		Crippen Method
mcvol	186.810	ml/mol	McGowan Method
pc	2370.28	kPa	Joback Method
rinqol	1656.00		NIST Webbook
tb	660.66	K	Joback Method
tc	883.48	K	Joback Method
tf	380.29	K	Joback Method
vc	0.697	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	508.95	J/molxK	660.66	Joback Method
cpg	584.76	J/molxK	846.34	Joback Method
cpg	571.76	J/molxK	809.20	Joback Method
cpg	557.72	J/molxK	772.07	Joback Method
cpg	542.59	J/molxK	734.93	Joback Method
cpg	526.34	J/molxK	697.80	Joback Method
cpg	596.75	J/molxK	883.48	Joback Method
dvisc	0.0002106	Paxs	660.66	Joback Method
dvisc	0.0002602	Paxs	613.93	Joback Method

dvisc	0.0003330	Paxs	567.20	Joback Method
dvisc	0.0004453	Paxs	520.48	Joback Method
dvisc	0.0006307	Paxs	473.75	Joback Method
dvisc	0.0009639	Paxs	427.02	Joback Method
dvisc	0.0016351	Paxs	380.29	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299075&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299075&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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