

Ethylene glycol bis(3-aminopropyl) ether

Other names:	1-Propanamine, 3,3'-[1,2-ethanediylbis(oxy)]bis-3,3'-(Ethylenedioxy)bispropylamine Propylamine, 3,3'-(ethylenedioxy)bis-Di(3-aminopropoxy)ethane
Inchi:	InChI=1S/C8H20N2O2/c9-3-1-5-11-7-8-12-6-2-4-10/h1-10H2
InchiKey:	POTQBGGSWSMCMX-UHFFFAOYSA-N
Formula:	C8H20N2O2
SMILES:	NCCCOCCOCCCN
Mol. weight [g/mol]:	176.26
CAS:	2997-01-5

Physical Properties

Property code	Value	Unit	Source
gf	-60.62	kJ/mol	Joback Method
hf	-405.31	kJ/mol	Joback Method
hfus	29.25	kJ/mol	Joback Method
hvap	59.50	kJ/mol	Joback Method
log10ws	-0.21		Crippen Method
logp	-0.283		Crippen Method
mcvol	155.280	ml/mol	McGowan Method
pc	2752.67	kPa	Joback Method
tb	572.34	K	Joback Method
tc	758.31	K	Joback Method
tf	390.90	K	Joback Method
vc	0.578	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	404.11	J/molxK	572.34	Joback Method
cpg	417.53	J/molxK	603.34	Joback Method
cpg	430.40	J/molxK	634.33	Joback Method
cpg	442.72	J/molxK	665.33	Joback Method
cpg	454.50	J/molxK	696.32	Joback Method

cpg	465.74	J/mol×K	727.32	Joback Method
cpg	476.43	J/mol×K	758.31	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2997015&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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