

# Benzeneacetic acid, 4-hydroxy-, ethyl ester

<b>Other names:</b>	Acetic acid, (p-hydroxyphenyl)-, ethyl ester Ethyl (p-hydroxyphenyl)acetate Ethyl 4-hydroxyphenylacetate 4-Hydroxyphenylacetic acid ethyl ester
<b>Inchi:</b>	InChI=1S/C10H12O3/c1-2-13-10(12)7-8-3-5-9(11)6-4-8/h3-6,11H,2,7H2,1H3
<b>InchiKey:</b>	HYUPPKVFCGIMDB-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O3
<b>SMILES:</b>	CCOC(=O)Cc1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	180.20
<b>CAS:</b>	17138-28-2

## Physical Properties

Property code	Value	Unit	Source
gf	-242.81	kJ/mol	Joback Method
hf	-435.31	kJ/mol	Joback Method
hfus	24.27	kJ/mol	Joback Method
hvap	62.30	kJ/mol	Joback Method
log10ws	-1.52		Crippen Method
logp	1.498		Crippen Method
mcvol	141.310	ml/mol	McGowan Method
pc	3664.21	kPa	Joback Method
rinpol	1559.00		NIST Webbook
rinpol	1559.00		NIST Webbook
tb	611.79	K	Joback Method
tc	835.24	K	Joback Method
tf	412.76	K	Joback Method
vc	0.477	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.48	J/mol×K	611.79	Joback Method
cpg	362.58	J/mol×K	649.03	Joback Method
cpg	373.90	J/mol×K	686.27	Joback Method

cpg	384.49	J/molxK	723.51	Joback Method
cpg	394.42	J/molxK	760.76	Joback Method
cpg	403.75	J/molxK	798.00	Joback Method
cpg	412.55	J/molxK	835.24	Joback Method
dvisc	0.0008548	Paxs	412.76	Joback Method
dvisc	0.0004034	Paxs	445.93	Joback Method
dvisc	0.0002113	Paxs	479.10	Joback Method
dvisc	0.0001203	Paxs	512.27	Joback Method
dvisc	0.0000734	Paxs	545.45	Joback Method
dvisc	0.0000473	Paxs	578.62	Joback Method
dvisc	0.0000320	Paxs	611.79	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17138282&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17138282&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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