

Isophthalic acid, ethyl 1-isopropyl-2-methylpropyl ester

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|-----------------------------|---|
| Inchi: | InChI=1S/C17H24O4/c1-6-20-16(18)13-8-7-9-14(10-13)17(19)21-15(11(2)3)12(4)5/h7-12 |
| InchiKey: | MPJHXBDJBWTCBO-UHFFFAOYSA-N |
| Formula: | C17H24O4 |
| SMILES: | CCOC(=O)c1cccc(C(=O)OC(C(C)C)C(C)C)c1 |
| Mol. weight [g/mol]: | 292.37 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -280.12 | kJ/mol | Joback Method |
| hf | -674.59 | kJ/mol | Joback Method |
| hfus | 28.44 | kJ/mol | Joback Method |
| hvap | 73.52 | kJ/mol | Joback Method |
| log10ws | -4.52 | | Crippen Method |
| logp | 3.701 | | Crippen Method |
| mvol | 241.510 | ml/mol | McGowan Method |
| pc | 1710.36 | kPa | Joback Method |
| rinpol | 1962.00 | | NIST Webbook |
| tb | 771.28 | K | Joback Method |
| tc | 980.14 | K | Joback Method |
| tf | 419.61 | K | Joback Method |
| vc | 0.909 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 705.98 | J/molxK | 771.28 | Joback Method |
| cpg | 722.05 | J/molxK | 806.09 | Joback Method |
| cpg | 737.00 | J/molxK | 840.90 | Joback Method |
| cpg | 750.84 | J/molxK | 875.71 | Joback Method |
| cpg | 763.59 | J/molxK | 910.52 | Joback Method |
| cpg | 775.27 | J/molxK | 945.33 | Joback Method |
| cpg | 785.88 | J/molxK | 980.14 | Joback Method |
| dvisc | 0.0012619 | Paxs | 419.61 | Joback Method |
| dvisc | 0.0005663 | Paxs | 478.22 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0003027 | Paxs | 536.83 | Joback Method |
| dvisc | 0.0001830 | Paxs | 595.45 | Joback Method |
| dvisc | 0.0001211 | Paxs | 654.06 | Joback Method |
| dvisc | 0.0000858 | Paxs | 712.67 | Joback Method |
| dvisc | 0.0000640 | Paxs | 771.28 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U356370&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/40-086-2/Isophthalic-acid-ethyl-1-isopropyl-2-methylpropyl-ester.pdf>

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