

# 4-Ethoxybenzoyl chloride

<b>Other names:</b>	p-Ethoxybenzoyl chloride Benzoyl chloride, 4-ethoxy-
<b>Inchi:</b>	InChI=1S/C9H9ClO2/c1-2-12-8-5-3-7(4-6-8)9(10)11/h3-6H,2H2,1H3
<b>InchiKey:</b>	XLWQUESMILVIPR-UHFFFAOYSA-N
<b>Formula:</b>	C9H9ClO2
<b>SMILES:</b>	CCOc1ccc(C(=O)Cl)cc1
<b>Mol. weight [g/mol]:</b>	184.62
<b>CAS:</b>	16331-46-7

## Physical Properties

Property code	Value	Unit	Source
gf	-118.17	kJ/mol	Joback Method
hf	-264.57	kJ/mol	Joback Method
hfus	19.70	kJ/mol	Joback Method
hvap	52.11	kJ/mol	Joback Method
log10ws	-2.90		Crippen Method
logp	2.464		Crippen Method
mvol	133.590	ml/mol	McGowan Method
pc	3239.34	kPa	Joback Method
tb	550.70	K	Joback Method
tc	772.80	K	Joback Method
tf	332.21	K	Joback Method
vc	0.504	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	280.82	J/molxK	550.70	Joback Method
cpg	292.35	J/molxK	587.72	Joback Method
cpg	303.20	J/molxK	624.73	Joback Method
cpg	313.39	J/molxK	661.75	Joback Method
cpg	322.92	J/molxK	698.76	Joback Method
cpg	331.82	J/molxK	735.78	Joback Method
cpg	340.09	J/molxK	772.80	Joback Method

dvisc	0.0016956	Paxs	332.21	Joback Method
dvisc	0.0010290	Paxs	368.63	Joback Method
dvisc	0.0006831	Paxs	405.04	Joback Method
dvisc	0.0004852	Paxs	441.46	Joback Method
dvisc	0.0003631	Paxs	477.87	Joback Method
dvisc	0.0002831	Paxs	514.29	Joback Method
dvisc	0.0002281	Paxs	550.70	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16331467&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16331467&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/40-094-3/4-Ethoxybenzoyl-chloride.pdf>

Generated by Cheméo on 2024-04-24 21:20:51.197210644 +0000 UTC m=+16282900.117787966.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.