

# Benzenemethanol, «alpha»-ethynyl-

<b>Other names:</b>	Benzy alcohol, «alpha»-ethynyl- «alpha»-Ethynylbenzyl alcohol «alpha»-Phenylpropargyl alcohol Phenylethynylcarbinol 1-Phenylpropargyl alcohol 2-Propyn-1-ol, 1-phenyl- 1-Phenyl-2-propyn-1-ol 1-Phenyl-2-propyne-1-ol Ethynylphenylcarbinol NSC 4326 1-phenylprop-2-yn-1-ol
<b>Inchi:</b>	InChI=1S/C9H8O/c1-2-9(10)8-6-4-3-5-7-8/h1,3-7,9-10H
<b>InchiKey:</b>	UIGLAZDLBZDVBL-UHFFFAOYSA-N
<b>Formula:</b>	C9H8O
<b>SMILES:</b>	C#CC(O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	132.16
<b>CAS:</b>	4187-87-5

## Physical Properties

Property code	Value	Unit	Source
gf	221.12	kJ/mol	Joback Method
hf	141.83	kJ/mol	Joback Method
hfus	16.65	kJ/mol	Joback Method
hvap	54.05	kJ/mol	Joback Method
ie	10.69	eV	NIST Webbook
log10ws	-2.21		Crippen Method
logp	1.353		Crippen Method
mcvol	111.180	ml/mol	McGowan Method
pc	4426.72	kPa	Joback Method
tb	513.86	K	Joback Method
tc	727.25	K	Joback Method
tf	310.40	K	Joback Method
vc	0.406	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	232.40	J/mol×K	513.86	Joback Method
cpg	242.80	J/mol×K	549.42	Joback Method
cpg	252.49	J/mol×K	584.99	Joback Method
cpg	261.52	J/mol×K	620.55	Joback Method
cpg	269.91	J/mol×K	656.12	Joback Method
cpg	277.71	J/mol×K	691.68	Joback Method
cpg	284.97	J/mol×K	727.25	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	388.00 ± 1.00	K	1.90	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4187875&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4187875&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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