

# Methylenecyclooctane

<b>Other names:</b>	Cyclooctane, methylene-
<b>Inchi:</b>	InChI=1S/C9H16/c1-9-7-5-3-2-4-6-8-9/h1-8H2
<b>InchiKey:</b>	GQRWNDZUODCEAJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H16
<b>SMILES:</b>	C=C1CCCCCCC1
<b>Mol. weight [g/mol]:</b>	124.22
<b>CAS:</b>	3618-18-6

## Physical Properties

Property code	Value	Unit	Source
gf	85.94	kJ/mol	Joback Method
hf	-82.51	kJ/mol	Joback Method
hfus	4.47	kJ/mol	Joback Method
hvap	36.87	kJ/mol	Joback Method
ie	8.79	eV	NIST Webbook
log10ws	-3.34		Crippen Method
logp	3.287		Crippen Method
mcvol	122.510	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
tb	428.00 ± 1.00	K	NIST Webbook
tc	655.42	K	Joback Method
tf	209.45	K	Joback Method
vc	0.442	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	239.94	J/molxK	437.24	Joback Method
cpg	325.04	J/molxK	619.06	Joback Method
cpg	309.84	J/molxK	582.69	Joback Method
cpg	293.74	J/molxK	546.33	Joback Method
cpg	276.73	J/molxK	509.97	Joback Method
cpg	258.80	J/molxK	473.60	Joback Method
cpg	339.36	J/molxK	655.42	Joback Method

dvisc	0.0002104	Paxs	437.24	Joback Method
dvisc	0.0003114	Paxs	399.27	Joback Method
dvisc	0.0005007	Paxs	361.31	Joback Method
dvisc	0.0008999	Paxs	323.35	Joback Method
dvisc	0.0018905	Paxs	285.38	Joback Method
dvisc	0.0049874	Paxs	247.41	Joback Method
dvisc	0.0187026	Paxs	209.45	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3618186&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3618186&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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