

# Decyl sulfide

<b>Other names:</b>	11-Thiaheneicosane Decane, 1,1'-thiobis- Didecyl sulfide di-n-Decyl sulfide didecyl sulphide n-Decyl sulfide
<b>Inchi:</b>	InChI=1S/C20H42S/c1-3-5-7-9-11-13-15-17-19-21-20-18-16-14-12-10-8-6-4-2/h3-20H2,1
<b>InchiKey:</b>	RKYMVQJWYYOIJU-UHFFFAOYSA-N
<b>Formula:</b>	C20H42S
<b>SMILES:</b>	CCCCCCCCCSCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	314.61
<b>CAS:</b>	693-83-4

## Physical Properties

Property code	Value	Unit	Source
gf	150.64	kJ/mol	Joback Method
hf	-414.26	kJ/mol	Joback Method
hfus	51.69	kJ/mol	Joback Method
hvap	66.93	kJ/mol	Joback Method
log10ws	-8.08		Crippen Method
logp	8.001		Crippen Method
mcvol	309.010	ml/mol	McGowan Method
pc	1026.63	kPa	Joback Method
rinpol	2295.00		NIST Webbook
rinpol	2295.00		NIST Webbook
rinpol	2285.00		NIST Webbook
rinpol	2281.00		NIST Webbook
tb	725.78	K	Joback Method
tc	900.07	K	Joback Method
tf	300.40 ± 2.00	K	NIST Webbook
vc	1.210	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	918.76	J/mol×K	725.78	Joback Method
cpg	939.38	J/mol×K	754.83	Joback Method
cpg	959.02	J/mol×K	783.88	Joback Method
cpg	977.73	J/mol×K	812.92	Joback Method
cpg	995.53	J/mol×K	841.97	Joback Method
cpg	1012.46	J/mol×K	871.02	Joback Method
cpg	1028.53	J/mol×K	900.07	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50751e+01
Coeff. B	-5.42985e+03
Coeff. C	-1.18144e+02
Temperature range (K), min.	485.34
Temperature range (K), max.	674.27

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C693834&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C693834&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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