

# Lauric acid, cyclohexylmethyl ester

<b>Inchi:</b>	InChI=1S/C19H36O2/c1-2-3-4-5-6-7-8-9-13-16-19(20)21-17-18-14-11-10-12-15-18/h18H
<b>InchiKey:</b>	VHTNBSNRTWIBSL-UHFFFAOYSA-N
<b>Formula:</b>	C19H36O2
<b>SMILES:</b>	CCCCCCCCCCCC(=O)OCC1CCCCC1
<b>Mol. weight [g/mol]:</b>	296.49

## Physical Properties

Property code	Value	Unit	Source
gf	-100.37	kJ/mol	Joback Method
hf	-625.97	kJ/mol	Joback Method
hfus	39.59	kJ/mol	Joback Method
hvap	67.47	kJ/mol	Joback Method
log10ws	-6.29		Crippen Method
logp	6.031		Crippen Method
mcvol	275.150	ml/mol	McGowan Method
pc	1283.75	kPa	Joback Method
rinsol	2163.00		NIST Webbook
tb	729.96	K	Joback Method
tc	916.79	K	Joback Method
tf	383.43	K	Joback Method
vc	1.056	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	846.17	J/molxK	729.96	Joback Method
cpg	867.17	J/molxK	761.10	Joback Method
cpg	887.03	J/molxK	792.24	Joback Method
cpg	905.76	J/molxK	823.37	Joback Method
cpg	923.39	J/molxK	854.51	Joback Method
cpg	939.96	J/molxK	885.65	Joback Method
cpg	955.50	J/molxK	916.79	Joback Method
dvisc	0.0021704	Paxs	383.43	Joback Method
dvisc	0.0008958	Paxs	441.19	Joback Method

dvisc	0.0004538	Paxs	498.94	Joback Method
dvisc	0.0002647	Paxs	556.69	Joback Method
dvisc	0.0001709	Paxs	614.45	Joback Method
dvisc	0.0001189	Paxs	672.20	Joback Method
dvisc	0.0000877	Paxs	729.96	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357925&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357925&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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